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APPLICATION OF FILM SPRAY TECHNOLOGY FOR THE ENHANCEMENT OF TRANSDERMAL DRUG DELIVERY SYSTEM- A REVIEW

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ABSTRACT

Conventional topical formulations are more beneficial for drug delivery systems and treatments, yet they hold a few drawbacks, including irritability, lower bioavailability of actives in sites, and poor local and systemic effects. In this present topical research, Film-Forming Sprays, falling under sprayed solutions, which would form a thin film coming into a contact with wound or open skin, acting as reservoirs for extended release of the active ingredient, like patches, but would ensure skin contours is the step to exceed the limitations of the conventional topical preparations. On contrast with conventional preparations, film-forming sprays have a prolonged release nature with uniform distribution of drugs and dosages, enhanced activities system availability, better penetrability, and ease of administration. This review describes the development of film-forming sprays in topical formulations containing several polymers and excipients that enhance the preparation properties and the stability of the active ingredients. This overview also summarizes the principal parameters in the preparation of film-forming sprays such as spray-ability, spread-ability, and film properties, in addition to the various types of polymeric concentrations, excipients, types of sprayers, and evaluations.

Keywords: Film-forming spray, Topical drug delivery, Polymer-based films, Wound healing, Controlled drug release, Mucoadhesion.

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INTRODUCTION

Topically applied drugs cause both local and systemic effects [1]. Topical efficacy of an administered medication depends upon the physicochemical properties of both the drug and excipients. Excipients support the formulations that allow them to stick to the skin layers for permeation and absorption by crossing the skin barriers and exhibiting therapeutic activity [2]. Skin exhibits low permeability to environmental compounds, its comparatively minimal surface area of 2 square meters, which allows it to receive around half of the blood flow throughout the body, makes it one of the most accessible organs in the human body. It acts as a defense against harmful toxins [3,4].

The external route or topical route provides high surface area with the additional advantage of convenient self-administration by potentially replacing hypodermic injection and oral medication delivery [2]. The stratum corneum is the outermost skin layer and the primary site of skin's percutaneous absorption, comprises sudoriferous or sweat glands, follicles, and hydrophilic and lipophilic domains. It is located in the level of keratin between the horny cells and has a 20% water content. The percentage of cross-linked keratin in the level of keratin between the horny cells is 20% [1,4].

The conventionally used topical dosage forms such as creams, patches, and ointments have various disadvantages such as hypersensitivity, irritation, scaling up in the production, uneven distribution, high dosing frequency, and poor application. A novel approach that can replace the conventional topical and transdermal formulations is the film-forming system (FFS) [4]. Is defined as a non-solid dosage form that when applied topically, forms a film on the skin. The formed film can be either a solid polymeric entity that serves as a matrix for the prolonged release of the drug into the skin or a residual liquid film that is quickly absorbed in the stratum [5-10].

Definition and mechanism of action of film-forming spray (FFS)

A film is formed in contact at the therapeutic location mediated by a sprayed solution drug delivery technique called an FFS [11,13,14].

The drug released at a controlled rate from the polymer matrix, almost in the same manner that a patch releases its contents after forming a film [1]. FFS is compared with the other topical applications; however, to create films in the regular structure of the skin or wound since the tiny droplets of the film-forming solution may reach deeply entrenched places. In this, the drugs will access the target tissue more readily. The dosages for the film-forming sprays can also be varied depending on the volume of solution in each spray to create localized or systemic effects. An FFS disperses uniformly and its user-friendliness can improve patient compliance [11,13-19]. Water spreads the thin film quickly [11,16]. This thin, smooth, non-sticky film improves the harsh, sticky texture of patches, ointments, gels, and other comparable items, in addition to being more pleasant for patients to move around on [17,18].

In addition, the thin film facilitates easier penetration to the wound moisture site or equilibrium maintenance. Just as patch preparations may cause discomfort or infection with poor patch preparation, poor wound humidity can do the same [19-21]. A special sprayer type is used to spray the solution forming droplets to create the film. Even all of these sprays have a unique set of specifications and applications; they are all unquestionably capacity of being used in medical field [21].

VARIOUS TYPES OF FORMULATIONS FOR FILM FORMATION

Sprays and solutions that form films are elegant and new concepts in the transdermal sustained-release drug delivery systems. The skin had smeared with a liquid or solution of the polymeric drug solution, which dries up to leave an almost transparent thin coating [9]. The four main ingredients in film-forming sprays and solutions are drugs, solvent systems, polymers, and penetration enhancers. The non-volatile component should prevent the drug from precipitating in solution when the volatile solvent component of the solvent system evaporates. The non-volatile ingredient is selected to disrupt the order of intercellular lipids and quickly separate into the stratum corneum, improving the diffusivity and skin penetration of the medication [7,8]. It also permeates the medication (drugs) distribution across the stratum corneum.

Table 1: Various types of film-forming sprayers (FFS) [22-26]

Spray specification	Features	References
Ordinary spray	 Usage of plastic or aluminum containers with a specification of 0.3 mm aperture and 1.2 mm diameter dip tube. No other specific or advanced technology is involved. It sprays 0.11-0.35 g/mL of film-forming solution at an angle of 78.69-87.39°. And leakage rate: 0.01-0.03%. Can be used in vertical or horizontal orientation. Horizontal spray nozzle maintains sterility during application and storage. Spray force depends on polymer 	[22-26]
Metered dosage spray	 type and concentration. Used for novel delivery like transdermal or transmucosal drug delivery. The amount of solution sprayed is crucial for drug dosage determination. Some factors affect the volume of spray: bottle capacity, container position, and homogeneity of particle dispersion. Normal spray duration is about 90–10 s. Spray angle is 83.51°. 	[22-26]
Electrostatic spray	 Leakage rate is about 0.01-0.02%. It is generally applied in agricultural sector primarily as an insecticide treatment. It also enhances spray efficiency, droplet formation rate, and coverage uniformity while reducing drift loss. Efficiency of spray depends upon the viscosity, surface tension, and electrical resistivity. Required conductivity is about 10⁸ to 10⁵ S/m. 	[22-26]
Ultrasonic spray	 Droplet diameter is about 6.3-26 µm. It produces nanoscale droplets with thin-film properties. It produces a uniform droplet<10 µm in diameter. Also operates in high and low pressure. Droplet diameter is about 1-10 µm. Nozzle diameter is about 0.5 mm. Uses a working electrode with a resonance frequency of 10 MHz. It is also used in medical applications such as layer-by-layer coating films. Also appropriate for utilizing a various source of polymers from natural and synthetic. 	[22-26]

There are three forms of film-forming formulas, they are,

- Film-forming spray/solutions
- Film-forming gel
- Film-forming emulsion.

Table 2: Various types of formulations used for film-forming

Category	Description	References
Film-forming spray/ solutions	Sprays and solutions that form films are a novel approach in transdermal sustained-release drug delivery. A polymeric drug solution was applied to the skin, and upon drying, it dispersed into a thin transparent film.	[9]
	Application: It dries to form a film, which is peel-able after treatment; it can used for bed return patients	[7-9]
	Example: Ketorolac film-forming polymeric solution using Eudragit in ethanol and PVP.	[9,10,27,28]
Film- forming gel	Gels are semi-solid systems containing liquid immobilized within a three-dimensional solid network, which can be hydrophobic or hydrophilic.	[29,57]
	Hydrogels: Hydrophilic polymers have been used as aqueous gels with water as the internal phase. Types: Two types of film-forming gels are available, they are Hydrophilic	[30]
	(hydrogels) and hydrophobic gels. Application: It can be used in thighs, stomach or arms, and shoulders, once it dries gel form a bioadhesive gel layer.	[31]
Film-forming Emulsion	Semi-solid or liquid form of film-forming emulsion that solubilizes both lipophilic and hydrophilic drugs. It is a mixture of oily and aqueous phases	[33]
	stabilized by emulsifying agents. Types of emulsions: Oil-in-water (O/W) and water-in-oil (W/O), depending upon the hydrophilic-lipophilic balance (HLB) of the emulsifier.	[33]
	Emulsifying agents: Polymers, surfactants, proteins (e.g., gelatine), finely divided solid particles (e.g., bentonite).	[34]
	Advantages: It covers a skin area with extended contact, making it suitable for chronic dermal therapy.	[35]

Mechanism involved in the film-forming spray

A thin, transparent film has left behind on the skin after the solvent evaporates from the Film-forming spray system. Because vehicle components are volatile and evaporate after being applied on the skin surface, thus the formula is greatly changed into a higher-strength film residue left on the skin surface. The drug concentration is increased in this stage, as it approaches supersaturation and it enhances the drug flux by increasing the skin's thermodynamic activity but without compromising its defense barrier. This formulation can thus, it can be used without any skin irritation or adverse effect [36,37].

The following explained modified version of Fick's law of diffusion discusses about the supersaturation. The following formula is describing about the Fick's law of diffusion barrier:

$$J = DKCv/h (1)$$

Where,

- J Indicates drug permeation rate (flux)
- $\ensuremath{\text{\textbf{D}}}$ Indicates drug diffusion coefficient
- Cv Indicates drug concentration
- h Represents the thickness of the barrier to diffusion.

Table 3: Components involved in film-forming spray

Components	Characteristics	References
Drug	Film-forming sprays are utilized to	[37,40,41]
	administer the hydrophobic and	
	hydrophilic drugs. Drugs with the	
	high strengths, are easily absorbed	
	drugs can be administered via the	
	skin are fairly painless.	
Polymer	Polymers can be used alone or in	[42-46]
	combination with other film-forming	
	polymers to deliver the desired film	
	properties. These polymers should	
	combine at skin temperature to form	
	a transparent, flexible film.	
	Polyvinyl pyrrolidine, hydroxypropyl	
	methylcellulose, chitosan, silicones,	
	and polydimethylsiloxane are some	
	examples of polymers.	
Solvents	Other categories of solvents were	[47-51]
	utilized while the film-forming	
	spray, solvents play a critical part in	
	dissolution as well as penetration of	
	the drug. The safety of such solvents	
	is questionable in the long term	
	due to their repeated utilization.	
	For example, benzyl alcohol,	
	butanol, propylene glycol, ethanol,	
	isopropanol, as well as ethyl acetate.	
Plasticizer	The plasticizer is mainly used in	[52-54]
	film-forming systems to provide	
	them with more tensile strength and	
	flexibility. Such plasticizers must	
	not diffuse quickly into the skin and	
	should be compatible with the used	
	polymers. Examples such as propylene	
	glycol, glycerine, polysorbate 80,	
	polyethylene glycol, sorbitol, triethyl	
	citrate, triacetin, and dibutyl phthalate	
	were used as a plasticizer.	
Permeation	Permeation enhancers are the	[54]
enhancer	compounds which are added in	
	transdermal drug delivery systems	
	(TDDS) that serves to enhance the	
	quantity of material (drug) which	
	can pass through the skin. Examples	
	of permeation enhancers include	
	alcohols such as ethanol, propylene	
	glycol, and isopropyl alcohol.	

The equation shows the drug concentration is directly proportional to the rate of penetration of the drugs to the skin. This, however, is the case when the drug has been dissolved completely in the vehicle. There is a clear and direct relationship between drug flux and the system's thermodynamic activity to saturation as shown in this equation [11]. The higher saturation, however, the higher the thermodynamic instability. FFS avoids this instability problem by creating a supersaturated system instantaneously after its application on the skin. Therefore, compared to other transdermal administration forms, it enhances the uptake of drugs through the skin [17,38].

This equation represents that the concentration of the drug and the rate at which it permeates the skin is directly proportionate. That is correct, though, if the medication has completely dissolved within the vehicle.

Equation 2 is the modified form of Fick's law of diffusion:

$$J = \alpha D/\gamma h \tag{2}$$

Where

 $\boldsymbol{\alpha}$ - Indicates the thermodynamic activity of the drug within the formulation

 γ - Indicates the thermodynamic activity of the drug within the membrane.

Therefore, according to this equation, the flow of drug relies directly on the saturation-related thermodynamic activity of the system. In contrast, thermodynamic instability increases with supersaturation [39].

EVALUATION PARAMETERS INVOLVED IN FFS [13,55,56]

Organoleptic parameters

Testing and observation ranged from color, odor, appearance, and many other physical attributes that described the FFS.

рH

The pH of the solution was measured using a digital pH meter. Before measuring the pH of several data sets, the pH meter had calibrated using phosphate support pH values of 4.0, 7.0, and 10. It was measured by immersing the pH meter's cathode into 20 mL of shower solution for a short period. The pH for each detailing was calculated a few times, and average attributes were considered.

Viscosity

The viscosity of the spray solution was determined using a Brookfield viscometer.

Surface tension and spray angle

The surface tension and spray angle of the formulation can be measured using a technique called Drop Count Method, where stalagmometer is used for same. The mathematical formulae to calculate were mentioned below.

Surface tension(
$$\sigma$$
) = $\sigma 1 \times \rho 2 \text{ n} 1/\rho 1 \text{ n} 2$ (3)

Where,

 σ and σ 1-denotes the Surface tension of the test liquid and the reference liquid usually, water, respectively

 $\rho 1$ and $\rho 2$ denote the density of reference liquids 1 and 2, respectively n1 denotes the number of drops of reference fluid falling from point A to B n2 denotes the number of drops of topical film-forming spray (test) solution falling from point A to B.

Spray angle (
$$\theta$$
) = tan-1 ($1/r$) (4)

Where,

 θ denotes the angle,

l denotes the distance between the sprayer and the paper,

r denotes the radius a sprayed circle.

Evaporation time/film formation time

The time it took for the film to completely dry, or evaporation time, was noted. How many solutions would have been sprayed onto the glass surface and how much time it would take for the film to become thoroughly dried.

Stickiness of film

The film adhesion was tested using fabric fibers, which were pressed against the film with minimal force and found to be gripped by the film. The degree of stickiness was rated as high, medium, and low.

Stability studies

Stability analyses of five sprays were formulated and preserved for 3 months. The formulations were filled in close-lid APF containers resting of ICH recommendations for the below conditions: $30^{\circ}\text{C}\pm2^{\circ}\text{C}/75\pm5\%$ RH. For initial data, samples from all sprays were collected at 0 and 3 months for assessment on the characteristics of appearance, pH, and viscosity.

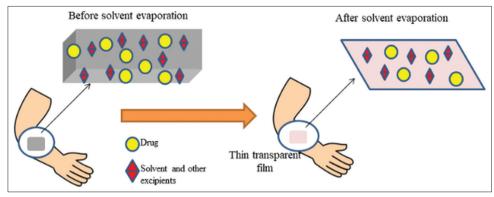


Fig. 1: Mechanism of film-forming system

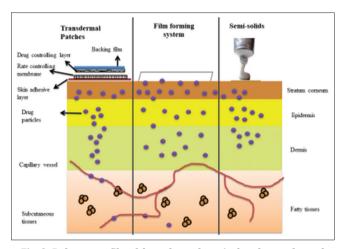


Fig. 2: Release profile of drug through topical and transdermal drug delivery system

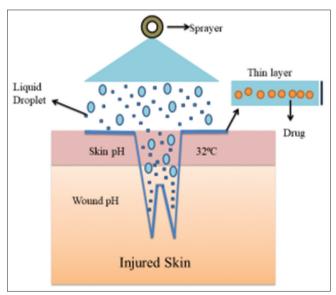


Fig. 3: Penetration of topical drug through skin

CONCLUSION

Therefore, it concludes that film-forming spray drug delivery system is a potentially helpful method of medication administration. Spray film-forming systems have been one of the modern drug delivery methods for local, topical, and transdermal administration. It is a unique and helpful technique offering a sustained-release drug delivery system

with increased resistance time, minimized skin irritation and dosing frequency, improved skin adhesion properties, increased drug release, and improved patient compliance. Polymers and excipients in film-forming sprays increase the stability of the active ingredients and the properties of the preparations. Films with different properties will result from different combinations between polymer and excipient. Therefore, all types of polymers and excipients would thus be incorporated into the composition of the film-forming spray, and the parameters by which they would be evaluated, and also thoroughly examined regarding the formulation properties.

AUTHORS' CONTRIBUTIONS

Mr. Jana Arun: Writing – review and editing, designed the review. Dr. GNK. Ganesh: Writing – review and editing, designed the review. Ms. Disha Bhattacharya: collected documents and literatures and Mr. Jagan Subramanian: Editing, designed the review.

COMPETING INTERESTS

The authors declare no conflicts of interest, financial or otherwise.

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ETHICS APPROVAL AND CONSENT TO PARTICIPATE

Not required for this study.

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