

## GREEN SYNTHESIS OF CHITOSAN-MEDIATED SILVER NANOCOMPOSITE USING *PIPER BETEL* STEM EXTRACT: EVALUATING ANTIBACTERIAL AND ANTICANCER ACTIVITY

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### ABSTRACT

**Objective:** The integration of herb-based nanomedicine has gained significant attention, aiming to improve therapeutic outcomes while minimizing drug toxicity. *Piper betel* is a medicinal herb widely used in traditional therapy due to its potent anticancer, antimicrobial, antioxidant, and anti-inflammatory properties.

**Methods:** *P. betel* stem extract is used to synthesize chitosan-mediated silver nanocomposite (AgNC). Characterization was done using UV-visible spectrophotometry, Fourier transform infrared (FTIR), X-ray diffraction, dynamic light scattering with Zeta potential, and FESEM with energy dispersive X-ray. Its antibacterial activity was evaluated in *Staphylococcus aureus*, *Pseudomonas aeruginosa*, and *Escherichia coli*, as well as antioxidant diphenyl-2-picrylhydrazyl and anti-inflammatory assays (egg albumin and bovine serum albumin) were performed. Further, its anticancer activity was evaluated in A549 and Henrietta Lacks (HeLa) cell lines.

**Results:** AgNC exhibited a sharp surface plasmon resonance peak at 450 nm. FTIR analysis revealed various peaks. Scanning electron microscopy analysis revealed a soft-fibrous sheet-like morphology. In antibacterial activity, a significant zone of inhibition was observed against *P. aeruginosa* and *E. coli* (16 mm). In addition, AgNC exhibited inhibition of free radicals and protein denaturation. In anticancer activity, it showed greater activity against A549 (66.7±5.9 µg/mL) than HeLa (85.7±7.8 µg/mL), and reduced cell migration within 48 h at 80 µg/mL in A549 cell lines.

**Conclusion:** These findings highlight AgNCs as the most promising candidates for biomedical applications, particularly for antibacterial and anticancer therapy. However, further investigations are required to elucidate the precise molecular mechanisms of therapeutic efficacies and modifications to enhance their anticancer activity.

**Keywords:** Green synthesis, Chitosan, Silver nanocomposite, *Piper betel* stem, Anticancer, Antibacterial.

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### INTRODUCTION

Nanotechnology has become an innovative approach in biomedical applications in recent years, providing improved efficacy in the treatment of disease and drug delivery [1]. Recently, the medical sector has expressed greater interest in herbal medicines to achieve sustainable activity and less toxicity of drugs [2]. Among the several nanomaterials, silver nanoparticles (AgNPs) have gathered significant attention due to their broad-spectrum antimicrobial properties and promising applications in cancer therapeutics [3]. Conventionally, chemical and physical methods have been employed to synthesize AgNPs; however, these methods often involve toxic reagents, expensive processes, and hazardous by-products. As a sustainable alternative, green synthesis has gained attraction, leveraging natural plant extracts as reducing and stabilizing agents to minimize environmental impact [4]. The use of plant-based extracts in nanomaterial synthesis offers several advantages, such as biocompatibility, cost-effectiveness, and the presence of bioactive phytochemicals that enhance the therapeutic potential of the synthesized nanomaterials [5].

*Piper betel* is an evergreen dioecious plant commonly grown throughout India. It belongs to the family of *Piperaceae*, also known as Paan in India and other subcontinental nations. It is a well-known medicinal plant that demonstrated a wealth of bioactive compounds ( $\beta$ -caryophyllene, eugenol, hydroxychavicol, estragole, caryophyllene, 1,8-cineol,  $\alpha$ -pinene,  $\beta$ -pinene, and hydroxy catechol), which exhibit antimicrobial,

antifungal, anticancer, antioxidant, and anti-inflammatory activity [6]. These phytochemicals (alkaloids, flavonoids, coumarins, tannins, terpenoids, cardiac glycosides, and saponins), chief components, chavibetol, chavicol, and polyols, enhance capping and reducing nature, making *P. betel* an excellent source for the green synthesis of NPs [7].

Chitosan can be used with silver (Ag) NPs to enhance its antibacterial and anticancer activity [5,8]. Chitosan interacts with nanometals to produce nanocomposites. These delivery methods span from microparticles to NPs, composites, nano-fibers, and films. Metallic NPs with chitosan improve the release of loaded anticancer medications, allowing sustained release, lowering the required dose, and minimizing harm to healthy cells [9]. Chitosan produced from crustacean animal shell waste has been widely used as an antifungal, medicinal coating, wound healing, and antioxidant [10]. The benefits of chitosan were that it is nontoxic, cheap, biocompatible, degradable, and water-soluble. It has been used in drug delivery, advanced cancer therapy [11]. Chitosan NPs (CNPs) are promising versatile cationic polymeric NPs, due to their incredibly tiny size; CNPs have fascinating interaction and surface properties. It showed effective antimicrobial activity against the pathogens *Escherichia coli*, *Pseudomonas aeruginosa*, *Klebsiella pneumoniae*, and *Staphylococcus aureus* [12,13].

Cancer is one of the life-threatening and leading causes of death. It can affect various organs and tissues, leading to severe health complications and high mortality rates [14]. Traditional treatments for cancer, such as

chemotherapy, radiation therapy, and surgery, are expensive procedures and also often come with side effects, including toxicity to healthy cells, drug intolerance, and drug resistance in a few patients [15]. Due to these drawbacks, continuous research is being conducted to explore natural, inexpensive, anti-proliferative, and chemopreventive agents as alternatives to synthetic drugs, mainly aiming to achieve lower toxicity with fewer side effects [16,17]. Ultrafine NPs were created using green methods, which is an essential property for a variety of applications where a particular surface area is important [11].

This research demonstrates a sustainable and efficient method for creating silver nanocomposites using an environmentally friendly approach, involving *P. betel* stem extract and chitosan [18]. Their physicochemical properties and biological activities were evaluated with an emphasis on determining their anticancer efficacy against Henrietta Lacks (HeLa), A549, and Human Embryonic Kidney (HEK) cell lines. Their antibacterial potential against pathogenic bacterium (*S. aureus*, *P. aeruginosa*, and *E. coli*).

### Hypothesis of the study

We hypothesized that a chitosan-silver nano composite (Chitosan-AgNC) made using *P. betel* stem extract would exhibit antibacterial and anticancer properties.

### Study objectives

1. To synthesize and characterize Chitosan-AgNC using *P. betel* stem aqueous extract
2. To evaluate the antibacterial activity of Chitosan-AgNC against *E. coli* and *P. aeruginosa*, and *S. aureus* using agar well diffusion, cytoplasmic leakage, protein leakage, and antibiofilm assays
3. To assess the *in vitro* anticancer potential of Chitosan-AgNC against A549 and HeLa cell lines using 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) and cell migration assays in the A549 cell line.

## MATERIALS AND METHODS

### Materials

HeLa cervical cancer cell line, A549 (Human lung cancer), and HEK cell lines were procured from the National Centre for Cell Sciences, Pune, India. Dulbecco's modified Eagle's medium (DMEM) (DMEM, Cat; 11995065, Gibco), Fetal bovine serum (FBS) (FBS: Cat; 26140095, Gibco, US), 1% penicillin-streptomycin (P4333, Sigma-Aldrich, St. Louis, MO, USA), 0.5% Trypsin-Ethylenediaminetetraacetic acid (EDTA) (Cat; 15400054, Gibco), CyQUAN TMTT reagent (Cat; V13154, Invitrogen, Thermo Fisher Scientific, USA). For anti-bacterial assays, Nutrient agar, Muller-Hinton agar, and BHI broth were purchased from HiMedia Laboratories Private Limited, India. Amoxicillin (Alkem Laboratories) was used as a standard drug for anti-bacterial assays, and diclofenac for anti-inflammatory assays.

### Instruments/equipment

Ultraviolet (UV-visible) spectrophotometer (Shimadzu UV-1900i, Shimadzu Corporation, Japan), Fourier transform infrared (FTIR) (Spectrum-2, PerkinElmer, Singapore), X-ray diffraction (XRD) (ARL EQUINOX 3000, Thermo Fisher Scientific, USA), dynamic light scattering (DLS) with Zeta potential (Malvern Zetasizer, Massachusetts, US), Field emission scanning electron microscopy (FESEM) images with energy dispersive X-ray spectroscopy (EDX) Scanning Electron Microscope (ZEISS, Gemini 1, Sigma 300, Germany), Enzyme-Linked Immunosorbent Assay reader (iMark, Bio-Rad, Japan), Inverted Microscope (EXI-310 Trinocular with FL Illuminator-3, Accu-scope, NY, USA), CO<sub>2</sub> incubator (Hera Cell Vios160i LK, Thermo Scientific, Germany), Digital Antibiotic Zone reader (1941, Electronics, Haryana, India), and Centrifuge (Sorvall ST 8R, ThermoScientific, Germany).

### Collection of plant and authentication

Fresh *P. betel* plant (Fig. 1) was collected from Annaji Nagar Park, 2<sup>nd</sup> Main Road, (GPS coordinates: 13°02'26.2"N 80°11'20.3"E), KK Nagar, Chennai, India. The authentication was done by the Central Council for Research in Siddha (Voucher specimen Code: P31082404B, and

Botanical name: *Piper betle* L.), Ministry of AYUSH, Anna Government Hospital Campus, Arumbakkam, Chennai, India.

### Preparation of aqueous extract

The plant was rinsed thoroughly with distilled water, and the stem was separated from the plant and then cut into small pieces [19]. These finely cut pieces were dried at room temperature for 2 weeks, then transferred to a hot air oven at 37°C for 3 days. Then, the dried stem was powdered using a mortar and pestle. This fine powder was stored in an airtight container at room temperature for further use.

1% aqueous extract was prepared by dissolving 1 g of the above-mentioned dry powder in 100 mL of double-distilled water and boiling for 30 min at 60°C using a magnetic stirrer [20]. The extract was filtered through Whatman Filter paper no. 1. This filtrate was utilized for AgNP synthesis.

### Synthesis of AgNPs

20 mL of *P. betel* stem filtered aqueous extract was added dropwise to 0.01M AgNO<sub>3</sub> solution and subjected to stirring for 20 min, resulting in the formation of colloidal nano-silver [21]. Then, it was transferred to an orbital shaker at 37°C. The transition in color indicates the formation of NPs.

### Synthesis of Ag-nano composite

#### Preparation of 1% (w/v) chitosan solution

To prepare 1% chitosan solution, 0.5 g of chitosan powder was dissolved in 9 mL of Distilled water. To facilitate the degradation of chitosan, 1 mL of 85% (v/v) glacial acetic acid was added to the mixture [22]. Then, this solution was stirred thoroughly at 800 rpm for 5 h to achieve a homogeneous mixture.

#### Preparation of chitosan-mediated silver nanocomposite (AgNC)

Equal volumes of the prepared *P. betel* stem AgNPs solution and 1% chitosan solution were mixed in a 1:1 ratio [23]. It was stirred using a magnetic stirrer for 30 min. Then, transferred to an orbital shaker for 48 h for composite formation. The resulting nanocomposite (AgNC) was collected by centrifugation at 8,000 rpm for 10 min [12,24]. The pellet was collected and washed with distilled water several times, and the obtained product was heated at 70°C for 6 h [25]. The powder was used for further processing.

#### Characterization of AgNC

The formation of AgNPs and AgNP-chitosan composite (AgNC) was primarily confirmed by measuring  $\lambda$  max with a UV-visible spectrophotometer wavelength range between 200 and 700 nm [2]. The functional groups were determined by FTIR spectroscopy, crystalline structure was performed using XRD, particle size and surface charge were determined using DLS and Zeta potential [11]. The shape and surface morphology were investigated using Scanning Electron Microscopy (SEM) FESEM. Elemental analysis was done using EDX.

#### Antibacterial activity

The antibacterial potential of AgNC was evaluated by the Agar well-diffusion method against *S. aureus*, *P. aeruginosa*, and *E. coli*. The MHA plate surface was inoculated with bacterial culture by spreading microbial inoculum over the entire media. Then, 4 wells were created using a sterile 1 mL pipette tip, each well labeled with a concentration of AgNC and standard amoxicillin. 25, 50, and 100  $\mu$ g of AgNC, standard drug loaded into the wells, left for incubation at 37°C for 24 h, and evaluated for zone of inhibition by measuring zone around the well using antibiotic zone reader [8,26].

#### Minimum inhibitory concentration (MIC) and minimum bactericidal activity (MBC)

The bacterial organisms were inoculated in Nutrient broth overnight, then diluted with 0.85% sterile saline in BHI broth to get 0.5

McFarland's standard ( $1 \times 10^5$  Colony Forming Units/mL). Then, the incubation was continued for 4 h at 37°C, and broth, diluted with AgNC from 0.5 to 50 µg/mL (0.5, 1, 5, 10, 15, 20, 25, 30, 35, 40, 45, 50 µg/mL). The negative control (-) was carried out with sterile broth, whilst the positive control (+) remained untreated. Finally, microtubes were incubated at 37°C for 24 h. The lowest concentration with no apparent turbidity was identified as the MIC. The MBC was calculated by subculturing the MIC tube onto nutrient agar and observing the absence of bacterial colonies following an overnight incubation [27].

#### Cytoplasmic leakage assay

Bacterial strains were cultured in BHI broth and incubated overnight at 37°C in an orbital shaker [28]. Then, the culture was exposed to different concentrations of AgNC (25, 50, 100 µg/mL) and standard drug (amoxicillin) for 24 h, then the supernatant was collected by centrifugation at 6,000 rpm for 15 min, and analyzed spectrophotometrically at 595nm.

#### Protein leakage assay

Barford assay was performed to analyze the protein leakage by bacterial cells [29]. The *S. aureus*, *P. aeruginosa*, and *E. coli* were cultured in BHI broth for 24 h. Each organism was treated with 25 µL of 25, 50, and 100 µg/mL of AgNC and the standard drug in 1 mL of cultured broth, incubated at 37°C for 12 h. After treatment, the tubes were centrifuged at 6,000 rpm for 15 min, and the supernatant was collected. For each sample, 25 µL of the supernatant was mixed with 200 µL of reagent and kept in the dark for 30 min. The absorbance was measured at 595 nm.

#### Antioxidant assay

##### Diphenyl-2-picrylhydrazyl (DPPH)

The antioxidant property of AgNC was evaluated using the DPPH protocol by Dhanraj and Rajeshkumar 2021 with slight modification [3,30]. To prepare 0.1 mM DPPH reagent, 4 mg of DPPH was dissolved in 100 mL of 100% Methanol to make a stock solution. 100 µL of 25, 50, and 100 µg/mL AgNC were added to 3 mL of DPPH solution. The reaction mixture was incubated at room temperature in dark conditions for 20 min. The absorbance of each sample was measured at 517 nm using a UV-vis spectrophotometer. Ascorbic

acid was used as a standard. % of radical scavenging activity (RSA) calculated using below formula.

$$\text{Percentage of RSA} = \frac{\text{Abs of control} - \text{Abs of sample}}{\text{Abs of control}} \times 100$$

#### Anti-inflammatory assay

##### Bovine serum albumin (BSA) denaturation assay

A heat-induced protein denaturation assay was carried out to evaluate the anti-inflammatory activity described by Chandra *et al.* 2012 with minor modifications [31]. 100 µL various concentrations of AgNC (25, 50, 100 µg/mL), standard drug diclofenac sodium added to 1 mL of 1% BSA in phosphate-buffered saline (PBS) (pH 6.4), incubated at 37°C for 20 min, then the samples were shifted to a water bath at 70°C for 5min. Turbidity was measured at 660 nm after the cooling samples. The % inhibition of BSA denaturation was calculated using the following formula:

$$\text{Inhibition of protein denaturation (\%)} = \frac{\text{Abs of control} - \text{Abs of sample}}{\text{Abs of control}} \times 100$$

##### Egg albumin (EA) denaturation assay

An EA denaturation assay was performed to evaluate the anti-inflammatory activity of AgNC described by Ameena *et al.*, in 2023 [32]. 0.2 mL of 1% freshly prepared EA mixed with 2.8 mL of 1×PBS (pH 6.4), 2 mL of different concentrations of AgNC (25, 50, 100 µg/mL), and 2 mL of diclofenac sodium (positive control) mixed well and incubated at 37°C for 15 min. Then, it was shifted to a water bath to heat at 70°C for an additional 5 min to induce denaturation. The absorbance of denaturation was measured at 660 nm using a UV-visible spectrophotometer. The % inhibition of EA denaturation was calculated using the above formula.

##### Antibiofilm activity

Anti-biofilm activity of AgNC tested by crystal violet assay. *S. aureus*, *P. aeruginosa*, and *E. coli* colonies were inoculated in BHI broth and incubated in an orbital shaker at 37°C overnight. 200 µL culture incubated with different concentrations of AgNC in designated wells, and amoxicillin was used as a Standard in 96-well plates, incubated for 48 h at 28°C. The formed biofilm was washed 3 times using 1×PBS, fixation was done using methanol for 15 min, and stained with 0.1% crystal violet for 30 min; unbound stain was rinsed with distilled water 3–4 times. Biofilm solubilized in ethanol and acetone (1:9). The optical density was measured using a microplate reader at 590 nm [29,33].

#### Anticancer activity

##### Cell culture

A549, HeLa, and HEK cell lines were supplemented with DMEM high glucose, 10% FBS, and 1% penicillin-streptomycin, and cultured in a CO<sub>2</sub> incubator with 5% CO<sub>2</sub> and 95% humidity at 37°C. The cells were sub-cultured after reaching 70% confluence by detaching cells using

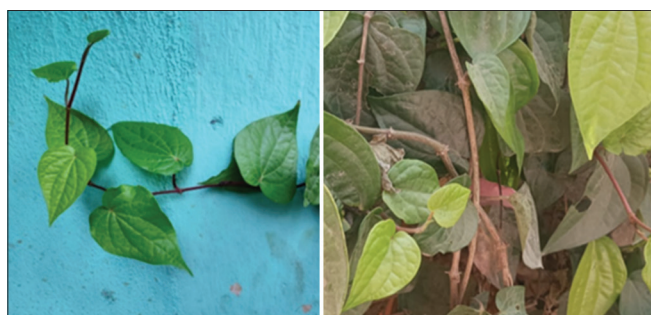


Fig. 1: Piper betel plant



Fig. 2: Visual observation of silver nanoparticles synthesis (a) AgNO<sub>3</sub> solution, (b) color change on the addition of piper betel stem aqueous extract, (c) silver nanoparticles with 1% chitosan

0.5% Trypsin-EDTA, and then the trypsin activity was deactivated by adding double the volume of complete medium.

#### Cell cytotoxicity (MTT)

Cell proliferation was quantified using the CyQUANT MTT reagent (Thermos Fisher), according to the protocol provided in product sheet [34]. The cells were seeded in 96-well plate ( $1 \times 10^4$  cells/well) and incubated to adhere the cells, then treated with cisplatin and AgNC (0, 10, 20, 40, 60, 80, 100, 200  $\mu\text{g/mL}$ ) for 24 h [35]. Upon treatment, the plates were replaced with 100  $\mu\text{L}$  of fresh media without phenol red indicator and incubated for another 4 h by adding 10  $\mu\text{L}$  MTT reagent per well. 100  $\mu\text{L}$  of SDS-HCL solution was added to each well mixed using a pipette, and incubated for 4–8 h. The absorbance was recorded at 570 nm. The viability percentage was calculated using the below formula,

$$\text{Percentage of cell viability} = \frac{\text{Abs of treated}}{\text{Abs of control}} \times 100$$

#### Cell migration assay

A stretch wound healing assay was done in A549 cell lines to assess Cell migration [36]. After getting 70% confluence, cells were seeded ( $2 \times 10^5$ /well) into a 6-well plate and cultured monolayer, and the wound was created in the middle of the plate using a sterile 200  $\mu\text{L}$  tip. Removed cell debris by washing with PBS and treated with AgNC (0, 20, 40, 80, 100  $\mu\text{g/mL}$ ) in complete media. Photos were captured at different time points (0, 24, and 48 h). The wound distance was evaluated using CaptaVision™ software (Accu-Scope, EXI- 310-FL3, USA). The percentage of wound closure is calculated using the formula;

$$\% \text{ of Wound closure} = \frac{W_0 - W_t}{W_0} \times 100$$

W<sub>0</sub>: Initial wound area; W<sub>t</sub>: Wound area after treatment at different time points.

#### Statistical analysis

All experiments were done in triplicate (n=3), and the data were analyzed by Two-way Analysis of Variance, followed by the difference between groups determined using Tukey multiple comparisons, and a Student's t-test used for two-group comparison. All statistical analysis was done using Prism GraphPad software version 10.4.1 (Windows). The data are presented as mean  $\pm$  standard error. Moreover, statistical significance is represented in asterisks (\*), \*p<0.05, \*\*p<0.01, \*\*\*p<0.001, \*\*\*\*p<0.0001, and ns indicates a p>0.05 statistically not significant.

## RESULTS AND DISCUSSION

#### Visual observation of NP formation

The formation of brown color was observed on addition of *P. betel* stem extract in 1 mM AgNO<sub>3</sub> within 3 h, and the intensity of color

strengthened during 24–48 h incubation (Fig. 2b) [37]. When it was mixed with Chitosan, it turned a light Greyish (Fig. 2c) [21]. Due to the presence of phytochemicals such as flavonoids, phenols, and alkaloids, they act as reducing and stabilizing agents [7]. Similar results were observed by Kanniah *et al.* in silver nanocomposite synthesis using *Piper nigrum* seed extract [4].

#### UV-vis spectrophotometer

The UV spectrum analysis of AgNPs synthesized from betel stem showed a maximum absorption peak at 450nm and declined absorption at higher wavelengths (Fig 3a). Indicating the presence of surface plasmon resonance of AgNPs and their formation [38,39]. The absorption spectrum analysis of AgNPs of betel stem extract on the addition of 1% Chitosan at 1 h slightly decreased in absorbance, indicating initial formation, at 18, 24, and 36 h further decrease in absorption indicates continued stabilization and possible changes in betel stem extract AgNPs with chitosan (Fig. 3b). This behavior aligns with the findings by (Arif *et al.*) [24], who reported similar trends in the interaction of AgNPs with stabilizing agents.

#### FTIR

The functional group's analysis of chitosan-based silver NP (AgNC) synthesized from betel stem extract exhibited major absorbance peaks at different wavelengths, indicating the presence of hydroxyl, amine groups, amide II, aliphatic chain, hydrocarbons, glycosidic, and metal-oxide bonds (Fig. 4). The broad peak at 3,255  $\text{cm}^{-1}$  corresponds to OH or NH functional groups and the role of chitosan as a reducing and stabilizing agent, as reported in the literature [40,41] in the synthesis of metal NPs. The peak at 1,544  $\text{cm}^{-1}$  and 1,535  $\text{cm}^{-1}$  related to N-H bending vibrations supports the involvement of amine groups in NP stabilization [39]. The peak at 1,403  $\text{cm}^{-1}$  corresponds to C-H binding vibrations, 1,013  $\text{cm}^{-1}$  I associated with glycosidic (C-O-C) bonds. Various peaks in a range between 400 and 700  $\text{cm}^{-1}$  attributed to metal-oxygen bonds indicate AgNP interaction with chitosan [5,8]. These results suggest the effective reduction and stabilization of AgNPs with chitosan.

#### XRD analysis

The crystalline nature of synthesized AgNPs and Ag-chitosan nanocomposite (AgNC) were characterized by XRD, and the resulting diffraction pattern is shown in Fig. 5. The AgNPs exhibited peaks at  $2\theta=38.28^\circ, 44.25^\circ, 64.46^\circ, 77.60^\circ,$  and  $87.52^\circ$ , corresponding to the planes of face-centered cubic (FCC) silver. The intense reflection at  $38.15^\circ$  indicates high crystallinity of AgNPs [42]. For AgNC diffraction peaks were observed at  $2\theta=21.56^\circ, 38.06^\circ, 44.25^\circ, 64.54^\circ, 77.55^\circ,$  and  $87.73^\circ$ . The broad peak centered around  $21.56^\circ$  corresponds to the semi-crystalline nature of chitosan, arising from the regular arrangement of polymeric chains. The distinct peaks appearing at  $38.06^\circ, 44.25^\circ, 64.54^\circ,$  and  $77.55^\circ$  are characteristic of FCC silver (Ag), and an additional weak peak at  $87.73^\circ$  may correspond to the metallic silver, confirming the successful formation of crystalline AgNPs within the chitosan matrix. These results were aligned with the previous

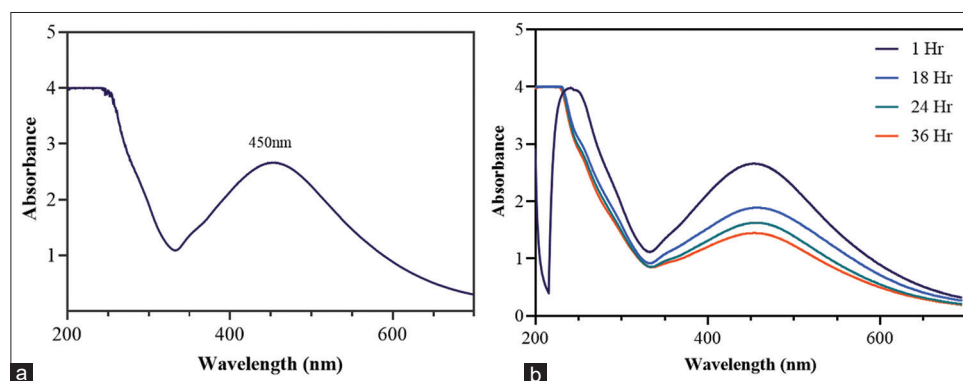
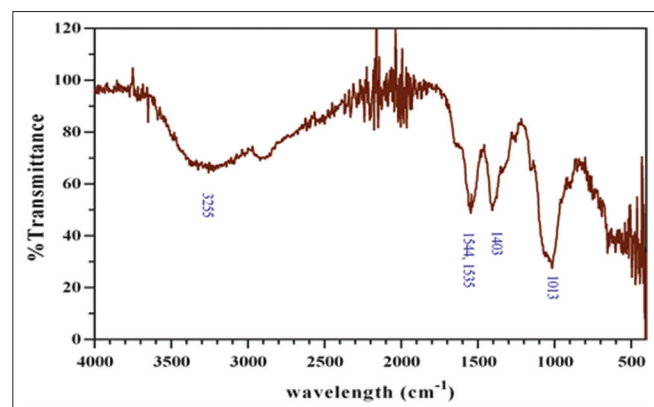


Fig. 3: Ultraviolet-visible spectrophotometer analysis of (a) silver nanoparticles, (b) silver nanocomposite

research done by Kalaivani *et al.* in 2018 and Mannopantar *et al.* in 2022 [11,40,43].

**DLS and zeta potential**

The successful green production and surface modification of AgNPs using chitosan and *P. betel* stem extract is confirmed by the DLS



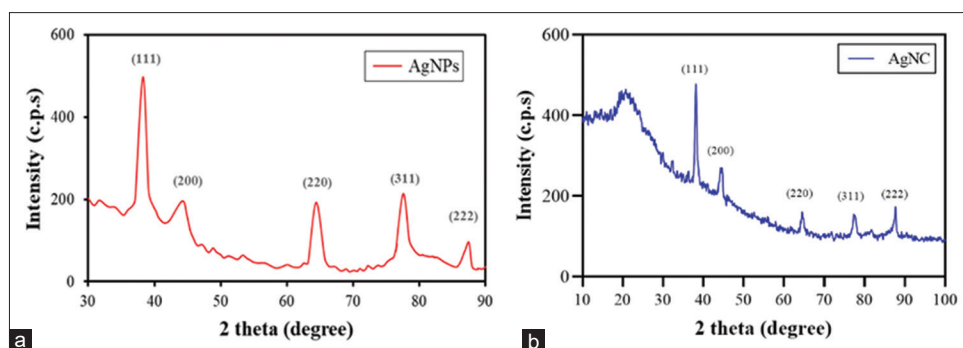
**Fig. 4: Fourier transform-infrared spectrum graph of silver nanocomposite, functional groups analysis**

and Zeta potential studies at temperature 25°C (Fig. 6). The AgNPs exhibited an average size of 138.8 nm and Zeta potential of -38.6 mV, while the silver nano composite (AgNC) displayed a large size 244 nm and higher negative surface charge -61.1 mV with polydispersity index (PDI)=0.403, these increase in particle size attributes the formation of chitosan coating layer, and the higher zeta potential indicates greater electrostatic stability (Tables 1 and 2), these results were correlated with Jalab *et al.* in 2021 the particle size and PDI was depends on concentration of extract and silver nitrate ratio [44].

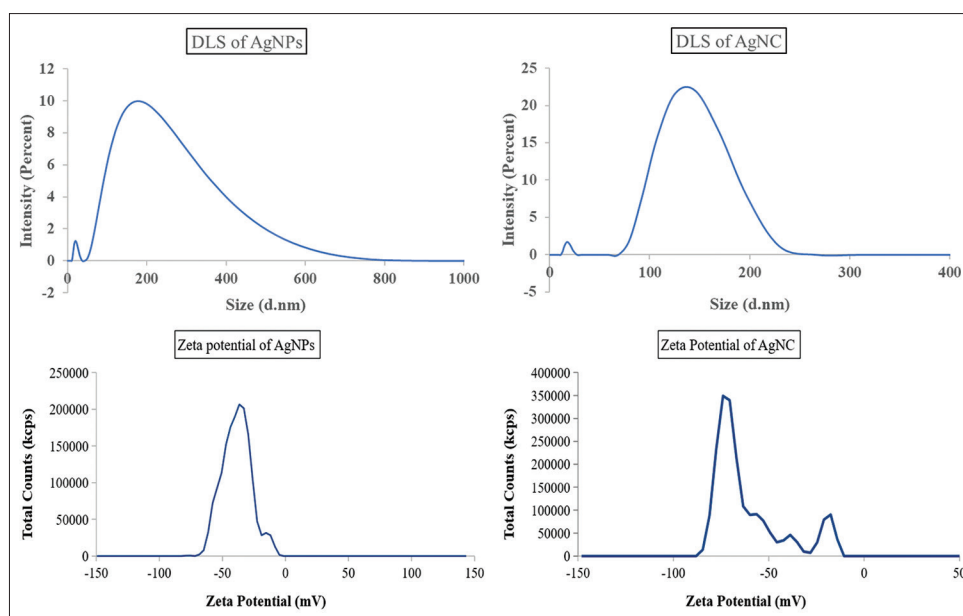
The low PDI (0.218) in nanocomposite supports enhanced homogeneity, colloidal stability, and a fairly monodispersed. The findings support *P. betel* stem extract efficacy as a bio-reductant and stabilizer, and chitosan as a biopolymer coating agent, which form stable AgNC with prospects of application in antimicrobial and biomedical fields [43,45].

**Morphological assessment (SEM)**

SEM imaging was used to investigate the morphology of AgNC at various magnifications, revealing distinct structural features. At lower magnification 10 μm, a layered non-homogeneous, non-smooth morphology with parallel ridges indicates structural alignment and possible integration of NPs (Fig. 7a) [4,39]. At 2 μm, the surface morphology with rough and porous structures in the micron range (Fig.7b) [18]. In higher magnification at 500 nm, it shows smooth, compact layers of closely packed structures, representing a chitosan



**Fig. 5: X-ray diffraction pattern of (a) silver nanoparticles, and (b) Ag-chitosan nanocomposite (silver nanocomposite) showing characteristic diffraction peaks of crystalline silver planes on the semi-crystalline chitosan background in silver nanocomposite**



**Fig. 6: Dynamic light scattering intensity and Zeta potential distribution of silver nanoparticles and Ag-Chitosan nanocomposite (silver nanocomposite) showing enhanced colloidal stability after surface modification**

matrix (Fig. 7c-d) with AgNPs [13]. These features of roughness and porosity indicate the successful incorporation of AgNPs with chitosan, enhancing drug-releasing efficiency.

#### Elements analysis EDX

The EDX spectrum revealed a Strong signal of silver (Ag) along with characteristic peaks of carbon (C) and oxygen (O), as well as minor peaks of Na, Al, Si, Cl, K, Ca, Fe, and Cu (Fig. 8a). The high carbon (44.26%) and oxygen (32.33%) corresponding to the chitosan polymeric framework and phytochemical capping agents derived from *P. betel* stem extract, as well as the prominent Ag peak, which confirmed the successful synthesis and presence of AgNP in the nanocomposite matrix (Fig. 8b and Table 3), accounting for 12.79%. These results are consistent with DLS and zeta potential studies, indicating that the nanocomposite is constituted of both metallic and biopolymeric phases, is well disseminated, and is chemically stable. The presence of Ag confirms NP formation, while C and O indicate involvement of chitosan and *Pipet betel* stem extract in the formation of the organic matrix.

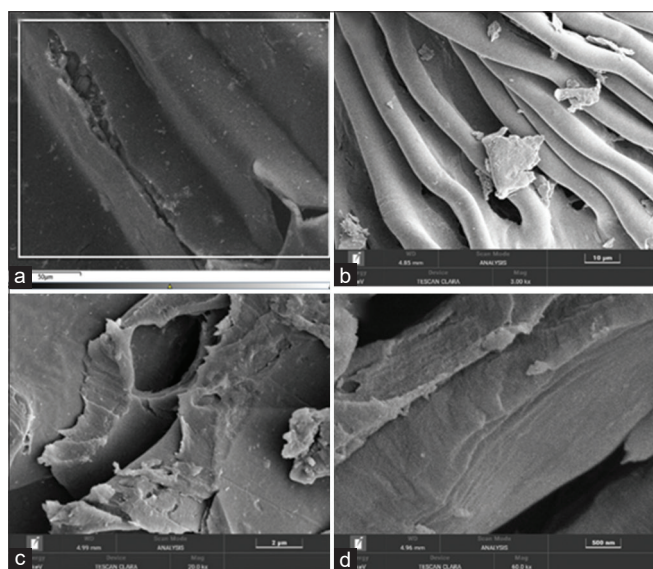


Fig. 7: Scanning electron microscopy images of silver nanocomposite; (a) 50  $\mu\text{m}$ , (b) 10  $\mu\text{m}$ , (c) 2  $\mu\text{m}$ , (d) 500 nm

#### Antibacterial assay

The AgNC exhibited concentration-dependent antibacterial activity against *S. aureus*, *P. aeruginosa*, and *E. coli* at concentrations of 25, 50,

Table 1: DLS parameters of AgNPs and AgNC nanocomposite

Parameter	AgNPs (silver nanoparticle) (%)	AgNC (silver nanocomposite) (%)
Z-average size (nm)	138.8	244
Peak 1 mean (nm) area (%)	210.6 (92.2)	138.6 (94.6)
Peak 2 mean (nm) area (%)	21.2 (5.16)	18.0 (5.3)
Polydispersity index	0.403	0.218

Table 2: Zeta potential parameters of AgNPs and AgNC nanocomposite

Parameter	AgNPs (silver nanoparticle)	AgNC (Silver nanocomposite)
Mean zeta potential (mV)	-38.64	-61.10
Zeta peak 1 (mV)	-76.87	-71.54
Zeta peak 2 (mV)	-39.83	-54.49
Conductivity (mS/cm)	1.781	0.437
Wall zeta potential (mV)	-38.17	-62.56
Quality factor	3.385	6.817
Temperature ( $^{\circ}\text{C}$ )	25	25

Table 3: Map sum spectrum of AgNC EDX elemental analysis with relative weight and atomic proportion

Element	Weight %	Atomic %
C	44.26	60.24
O	32.33	33.03
Na	1.29	0.91
Al	1.41	0.85
Si	0.13	0.08
Cl	0.88	0.41
K	3.20	1.34
Ca	1.47	0.60
Fe	0.77	0.23
Cu	1.49	0.38
Ag	12.79	1.94
Total	100.00	100.00

EDX: Energy dispersive X-ray

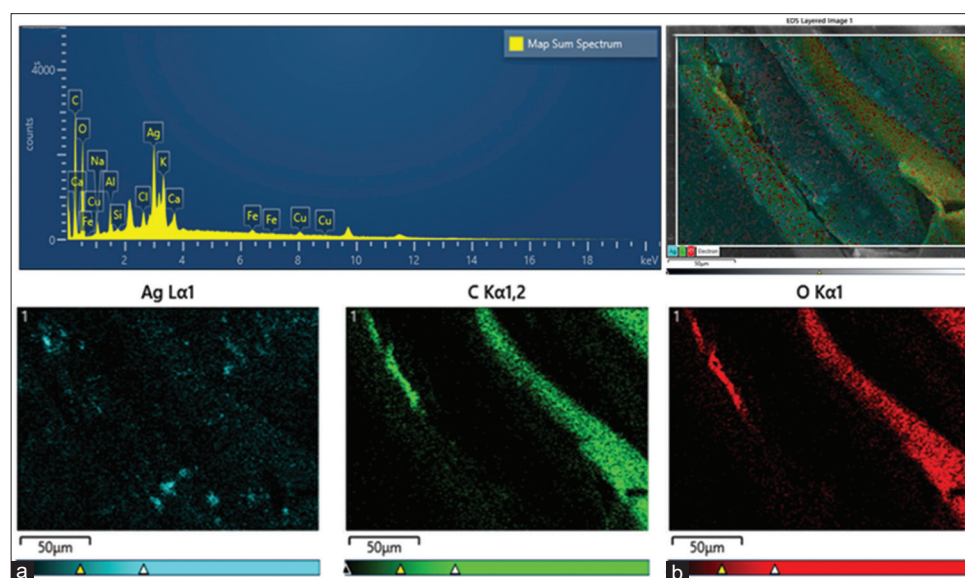


Fig. 8. Scanning electron microscopy analysis with energy-dispersive X-ray spectroscopy, (a) Shown an energy dispersive X-ray spectrum, (b) a corresponding elemental Map uniform distribution of silver (Ag, blue), carbon (C, green) and oxygen (O, red)



**Fig. 9: Antibacterial activity of silver nanocomposite in (a) *Staphylococcus aureus*, (b) *Pseudomonas aeruginosa*, and (c) *Escherichia coli* using agar well-diffusion method**

and 100 µg/mL. The zone of inhibition was measured in millimeters (mm), and the results were compared with a standard antibiotic (Fig. 9). For *S. aureus*, the zone of inhibition was 10 mm, 13 mm, and 14 mm, respectively, compared with the standard 16 mm. Against *P. aeruginosa*, inhibition zones were 12 mm, 15 mm, and 16 mm, demonstrating superior efficacy at higher concentrations compared to the standard antibiotic (15 mm). This result suggests that AgNCs may have better antibacterial activity against *P. aeruginosa*. For *E. coli*, the inhibition zones of the nanocomposite were 11 mm, 12 mm, and 16 mm, while the standard showed 14 mm; this indicates that AgNCs exhibit strong antibacterial efficacy [46,47]. The highest concentration (100 µg/mL) demonstrated a significant antibacterial activity across all bacterial strains (Fig. 10). These results correlated with previous studies, phytochemicals of *P. betel* extract alkaloids, eugenol, tannins and silver ions interact with esterase DNA, RNA, polymerase, and cell respiration play an important role in DNA intercalation acts as antibacterial agents [13,46].

#### Determination of MIC and MBC

The results revealed that the AgNC indicated strong growth inhibition and absence of turbidity at relatively low concentrations at 20 µg/mL for *S. aureus*, 15 µg/mL for *P. aeruginosa*, and *E. coli*. The MBC for all tested organisms was ranged between 30 and 40 µg/mL; these results demonstrate the potent antibacterial potential of biosynthesized Ag-chitosan nanocomposite (AgNC) [48].

#### Cytoplasmic leakage assay

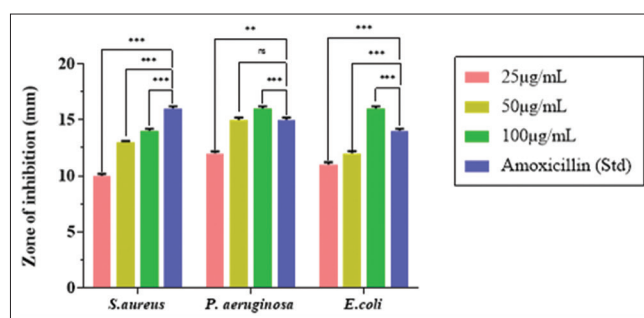
The concentration-dependent cytoplasmic leakage was observed in the treatment group compared with the positive control; higher concentrations exhibited more significant effects. *P. aeruginosa* and *E. coli* showed higher levels of cytoplasmic leakage compared to *S. aureus* (Fig. 11). Increased cytoplasmic leakage in *P. aeruginosa*, *E. coli* indicates the potent antibacterial effect of the AgNC, possibly due to the interaction of AgNC with bacterial proteins and DNA, leading to oxidative stress and cellular damage [10]. The interaction of AgNC with bacterial cell walls and the generation of reactive oxygen species (ROS) likely contribute to bacterial membrane disruption and cell death.

#### Protein leakage assay

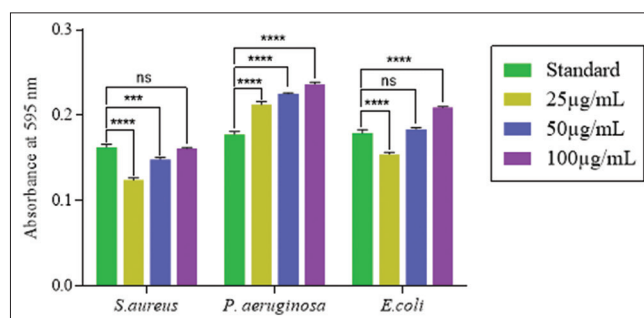
The protein leakage from cells was increased along with an increase in AgNC concentration, compared with standard (Fig. 12). A significant difference was observed with all concentrations in *P. aeruginosa* and *E. coli*. Maximum protein leakage was noted in 100 µg/mL. Moreover, Tannins are bioactive compounds that can coagulate bacterial protoplasm, cause proteins to precipitate and bind to one another, and inhibit the development of bacterial cell walls [8,33].

#### Ani-biofilm activity

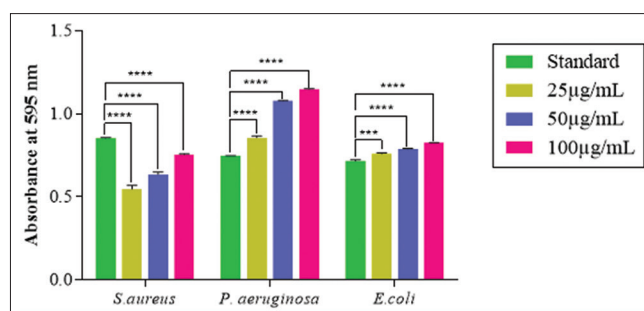
The results indicate a dose-dependent inhibition of biofilm formation; the control exhibited the highest biofilm mass, while treatment groups with increasing concentration significantly reduced biofilm formation (Fig. 13). Significant anti-biofilm activity was observed at higher concentrations (50–100 µg/mL), with *S. aureus* and *E. coli* showing



**Fig. 10: Antibacterial activity of betel stem silver nanocomposite. Results were presented in mean±standard deviation, n=3 (triplicates), \*\*p<0.01, \*\*\*p<0.001, nsp>0.05=not significant**



**Fig. 11: Cytoplasmic leakage analysis of silver nanocomposite. Results were presented in mean±standard deviation, n=3 (triplicates), \*\*\*p<0.001, \*\*\*\*p<0.0001, and ns p>0.05**

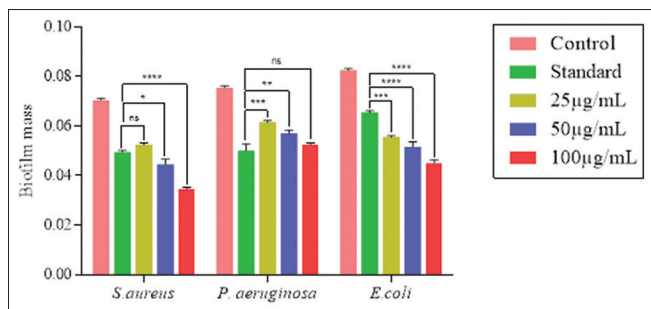


**Fig. 12: Protein leakage analysis of silver nanocomposite using Bradford reagent assay. Data presented in mean±standard deviation, n=3 (triplicates), two-way analysis of variance statistical significance was shown with asterisk (\*),\*\*\*p<0.001, \*\*\*\*p<0.0001, and ns p>0.05**

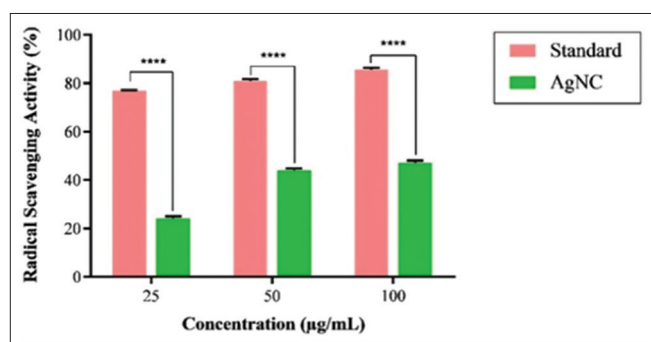
the strongest response. These findings align with earlier research that has reported antibiofilm activity of silver-based NPs and silver

nanocomposites that possess strong antibiofilm activity by disrupting bacterial cell walls and generating ROS [38,49]. Furthermore, higher silver concentration leads to enhanced penetration into the cell walls

and bacterial membrane damage [50,51]. Further research is needed to explore the mechanistic pathways of AgNC exerting biofilm-inhibitory effects.



**Fig. 13: Anti-biofilm activity of silver nanocomposite.** Values are expressed as the mean and standard deviation error bars of n=3 (replicated), statistical significance were presented with asterisk (\*), \*p<0.05, \*\*p<0.01, \*\*\*p<0.001, \*\*\*\*p<0.0001, and ns p>0.05



**Fig. 14: Anti-oxidant activity of silver nanocomposite using diphenyl-2-picrylhydrazyl assay.** Bar graph was presented with mean percentage and standard deviation error bars, n=3 (replicated), statistical significance was presented with asterisk (\*), \*\*\*\*p<0.0001

**Antioxidant assay**

The DPPH assay showed concentration-dependent free radical inhibition activity by AgNC NPs (Fig. 14). The percentage of antioxidant activity (24, 47, 44) increased with concentrations from 25, 50, and 100 µg/mL, which was lower compared to the standard. This was supported by Sonphakdi *et al.*, who reported that phenolic compounds donate electrons to other molecules and exhibit antioxidant activity [7]. These findings highlight the effectiveness of AgNC in reducing oxidative stress, which could be beneficial for biomedical and pharmaceutical applications.

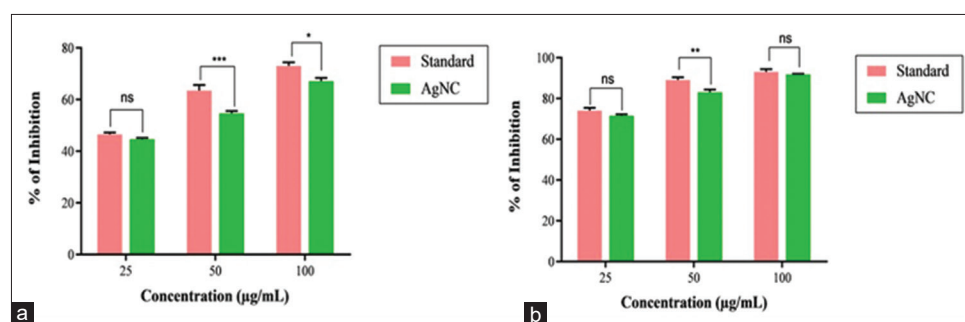
**Anti-inflammatory assay**

The anti-inflammatory property of AgNC in EA and BSA assays confirmed that it effectively inhibits protein denaturation, a key factor in inflammation. The percentage of inhibition was increased with an increase in concentration (Fig. 15), with the maximum inhibition observed at 100 µg/mL (67%, 92%). The EA assay showed that the inhibition was nearly equivalent to the standard drug. Similarly, the BSA was slightly less than the standard drug. These findings highlight the strong anti-inflammatory potential of AgNC, closely matching the effectiveness of the standard drug [32]. Moreover, reinforce the potential of AgNC in the management of inflammation. The bioactive components phenol compounds, flavonoids, and terpenoids are responsible for anti-inflammatory activity [6].

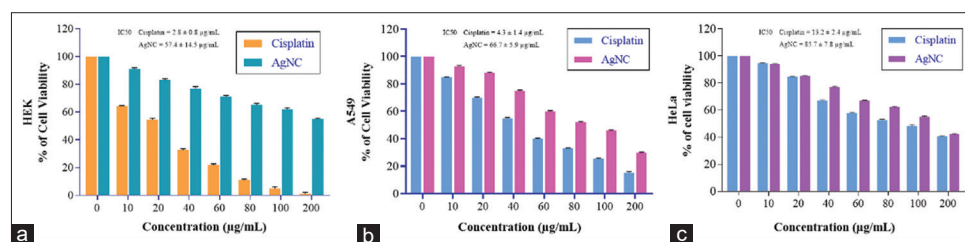
**Anticancer activity**

*Cytotoxicity assay (MTT)*

*In vitro*, cytotoxicity activity results of AgNC in A549, HEK, and HeLa cell lines were presented as a mean percentage of viability (Fig. 16). AgNC exhibited dose-dependent cytotoxicity, but its higher IC<sub>50</sub> suggests less potential than cisplatin in both cancer cell lines (Table 4). The greater cytotoxic effect was observed in A549 cells, suggesting higher susceptibility compared to HeLa cells with AgNC. Furthermore, we found less cytotoxicity towards HEK cell lines compared to the standard drug cisplatin [52,53]. The results suggest that AgNC has



**Fig. 15: Anti-inflammatory assay of silver nanocomposite, through the (a) egg albumin and (b) bovine serum albumin, bar graph was presented with mean percentage and standard deviation error bars, n=3 (replicated), \*p<0.05, \*\*p<0.01, \*\*\*p<0.001, \*\*\*\*p<0.0001, and ns p>0.05**



**Fig. 16: Effect of Piper betel silver nanocomposite and cisplatin at different concentrations in (a) human embryonic kidney, (b) A549, and (c) Henrietta Lacks cell lines.** Data represented in graph as viability mean with standard deviation error bars, n=3

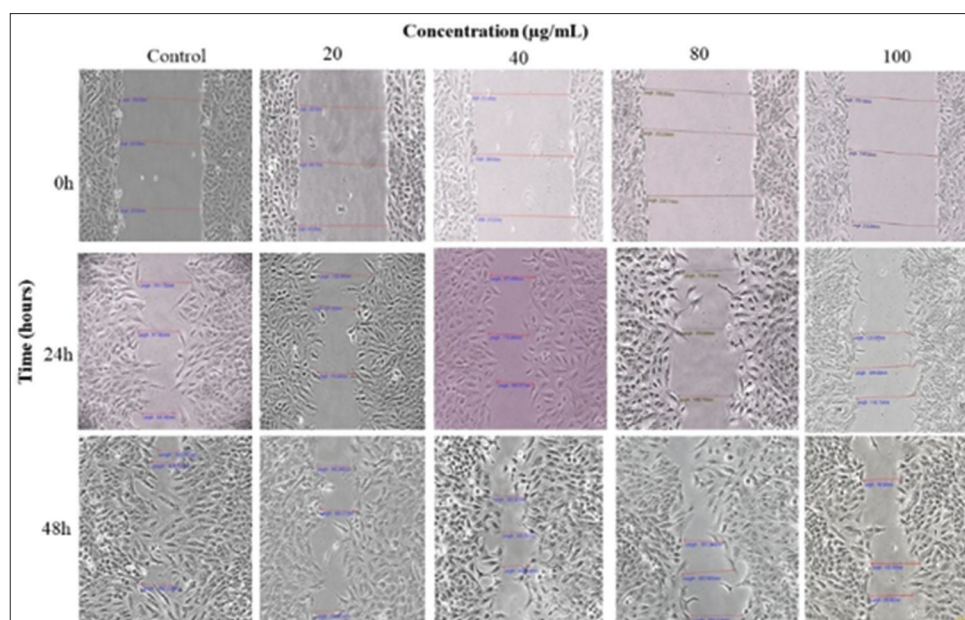


Fig. 17: Effect of different concentrations of silver nanocomposite in A549 cell lines migration at 0, 24, and 48 h

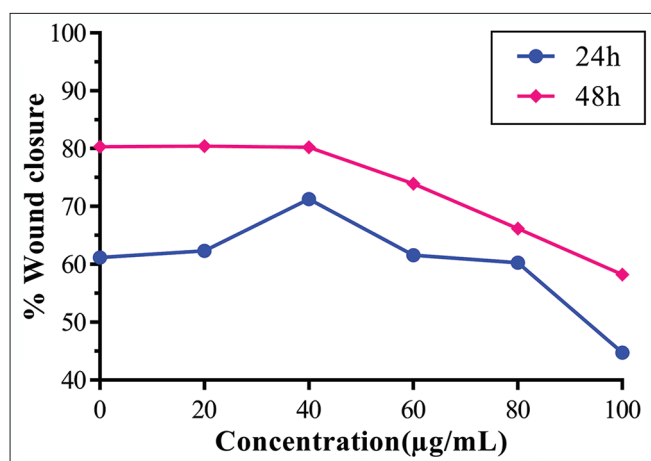


Fig. 18: Effect of silver nanocomposite on wound closure of A549 cell lines at 24 and 48 h, mean of wound closure are plotted in graph as percentage

potent anticancer activity but requires a higher concentration, and the potential of AgNC is possibly enhanced with the combination of other drugs for its effectiveness [22,54]. These results were correlated with previous research conducted in *P. betel* AgNPs, conjugation of AgNPs with chitosan [55-57]. Several studies have reported that cytotoxicity depends on several factors, such as the particle size of NPs and the type of cell line [58]. As a result, more investigation is required to use enhance the AgNC as an effective cancer therapy medicine.

#### Cell migration (scratch wound healing) assay

The cell migration assay exhibited significant differences at various concentrations across different time points (0 h, 24 h, 48 h) and drug concentrations (20, 40, 80, and 100 µg/mL) compared to the control group (Fig. 17). Cell migration reduction was observed at 24 h compared to 0 h. The significant decline of cell migration at 48 h indicates a substantial inhibition at 48 h. Overall, the results suggest a time-dependent decrease in cell migration from 0 to 48 h (Fig. 18), with an increasing concentration of AgNC. Notably, a significant inhibition was observed at 80 µg/mL (\*\*p=0.001) and 100 µg/mL (\*\*p<0.0001) [52].

Table 4: IC<sub>50</sub> values (µg/mL) of cisplatin and AgNC in HEK, A549, and HeLa cell lines

Cell line	Cisplatin	AgNC
	(Mean±SD)	(Mean±SD)
A549	4.3±1.4	66.7±5.9
HeLa	13.2±2.4	85.7±7.8
HEK	2.8±1.0	57.4±14.5

Values are expressed as mean±SD (standard deviation) from three independent experiments (n=3), units were in micrograms per milliliters (µg/mL)

#### CONCLUSION

In this present study, successfully synthesized chitosan-mediated silver nanocomposite (AgNC) using an aqueous *P. betel* stem extract. AgNC possessed significant antibacterial properties against pathogenic bacteria. Furthermore, it shows effective results for cytoplasmic leakage, protein leakage, antioxidant, and anti-inflammatory activity. Anticancer activity against the A549, HeLa cell lines. This study suggested that a combination of chitosan and AgNPs with green synthesis will be a good combination for developing an alternative therapeutic agent with fewer side effects and may provide a synergistic anticancer and antibacterial activity. Which may facilitate the development of a new class of antimicrobial and anticancer agents, further research is required on drug release kinetics and in *in vivo* toxicity evaluation to assess the biocompatibility and safety.

#### Limitations

The present study demonstrates cytotoxic and antimigratory effects but does not experimentally confirm the mechanism of cell death; assays such as Annexin V/PI and ROS analysis will be included in future studies.

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#### AUTHOR CONTRIBUTIONS

K.N.K., N.D., and P. AK contributed to conceptualization, methodology, review, and editing. M.S., M.B.S. performed the experiments and

collected the data. M.S., K.N.K., and N.D. drafted the manuscript, T.M and N.D. supervised the study and provided technical guidance. M.S., and N.K.K. performed the statistical analysis. All authors interpreted the results and critically reviewed and revised the final manuscript.

#### CONFLICT OF INTEREST

The authors declare that there is no conflict of interest regarding the publication of this article.

#### ETHICAL APPROVAL

No animal or human studies were involved.

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