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Original Article

DEVELOPMENT AND VALIDATION OF UV-SPECTROPHOTOMETRIC AND RP-HPLC METHODS FOR CURCUMIN-TOFACITINIB NANOCARRIERS: A NOVEL PLATFORM FOR ENHANCED BREAST CANCER THERAPY

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ABSTRACT

Objective: The objective of this study is to develop and validate analytical techniques for the accurate quantification, stability assessment, and quality control of curcumin-tofacitinib-loaded nanocarriers (CS-TF-NCs).

Methods: A new RP-HPLC (Reverse Phase-High Performance Liquid Chromatography) and UV-spectrophotometric method was developed and validated to quantify CS (Curcumin), TF (Tofacitinib), and CS-TF-NCs (Curcumin-Tofacitinib-Nanocarriers). UV-spectrophotometry detected λ max values of 426 nm (CS) and 286 nm (TF), and shifts to 287 nm and 421 nm for nanocarrier-loaded samples, with excellent linearity (R^2 = 0.9994) in the range of 5–25 μg/ml. RP-HPLC analysis was found to be very sensitive with low LOD (Limit of Detection) (LOD = 0.045 μg/ml) and LOQ (Limit of Quantification) (LOQ = 0.07 μg/ml). Recovery levels were between 97% and 99%, and retention times were 2.212 and 4.285 min for CS and TF, respectively.

Results: CS and TF revealed λ max values of 426 nm and 286 nm by UV-spectrophotometry, with high correlation coefficients of R^2 = 0.9962 and 0.998. For CS-TF-NCs, the values were determined at 287 nm and 421 nm with a high R^2 = 0.9994 that validates linearity for the entire range of concentration of 5–25 µg/ml. Validation through RP-HPLC has exhibited good sensitivity, LOD being 0.045 µg/ml, and high precision (%RSD<1.5%). The nanocarriers were characterized by good solubility and sustained release profiles, giving recovery and retention times of 97–99% with 2.212 and 4.285 min for CS and TF, respectively.

Conclusion: This work established CS-TF-NCs as a platform with the potential to deliver targeted, effective breast cancer therapy.

Keywords: Breast cancer, Curcumin, Tofacitinib, Nanocarriers, Combination therapy, RP-HPLC, UV-spectrophotometry, Drug solubility, Bioavailability, Stability studies

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INTRODUCTION

The cornerstone of modern anticancer therapy is a combination therapy that takes advantage of synergistic or additive effects conferred by two or more therapeutic drugs acting on various pathways in the tumour's growth [1, 2]. Combination therapy has been reported to suppress tumour growth and drug resistance, together with the ability to metastasize and reduce the number of cancer stem cells [2]. Of all the crucial pathways involved in breast cancer, the IL-6/JAK2/STAT3 (Interleukin 6/Janus Kinase 2/Signal Transducer and Activator of Transcription 3) pathway is the most pivotal pathway involved in tumorigenesis, proliferation, and metastasis [3]. Misregulation of the path has the potential to induce oncogenic signalling; hence, small-molecule inhibitors against the JAK/STAT proteins have been effective [4]. Curcumin (CS), a polyphenol, is a compound obtained from natural origins and possesses the ability to regulate the JAK/STAT pathway.

Analytical research on CS, for instance, by Zoiet al., has been reported to inhibit STAT3 phosphorylation and downregulate its downstream gene expression, which results in decreased tumour burden in breast cancer models [5]. Likewise, the effective JAK-STAT pathway inhibitor, tofacitinib (TF), drastically lowers the viability of breast cancer cells, augmenting the effectiveness of other anticancer drugs, according to research by Gravina et al., [6]. CS and TF are presently established to be novel therapeutic strategies in synergistic mixtures directed toward the IL-6/JAK/STAT3 pathway. Though CS is a potent therapeutic molecule, it is associated with low solubility, bioavailability, and stability [7].

TF is hampered by systemic toxicity and lacks selectivity. Thus, the research work presently focuses on mitigating these disadvantages with the help of nanocarrier-based drug delivery systems [8]. Al Thani *et al.*, showed that nanocarriers enhance the solubility and stability of the contained agents, which might allow for site-specific targeting at tumour sites and thereby reduce off-target effects [9].

More to that were the characteristics V. C. Deivayanai *et al.* showed for long-term drug delivery and therapeutic effects in cancer treatment. Quantitative analysis methods are vital in assessing the effectiveness of such nanocarrier systems [10].

Methods that Gorantla et al. came up with utilized UVspectrophotometric methods of drug quantitation in bulk materials. However, such methods cannot provide the sensitivity needed to undertake complex formulations using nanocarriers [11]. RP-HPLC methods for the study of stability in anticancer drugs such as TF and CS were developed by Srividya Gorantla et al. Yet, these methods tend to require enhancement in managing drug combinations in nanoscale systems [12]. The intricacy of curcumin-tofacitinib nanocarriers requires more stringent and accurate methodologies. This study will fill these lacunae through the design and validation of sophisticated analytical methods for curcumin-tofacitinib-loaded nanocarriers. Through this, it hopes to enhance UV-spectrophotometric and RP-HPLC methods for the reliable quantification and characterization of the drugs in both bulk and pharmaceutical states. In addition to overcoming significant shortcomings in earlier approaches, this research also offers a platform for the successful clinical application of curcumin-tofacitinib nanocarriers in the treatment of breast cancer.

MATERIALS AND METHODS

Chemicals and reagents

Curcumin (CS) was obtained from SRL Chemicals, and Tofacitinib (TF) was kindly provided as a gift sample by Hetero Pharma Ltd., India. HPLC-grade methanol was procured from Sigma Aldrich Corporation, Mumbai, provided. Other reagents used in the study include ortho-phosphoric acid, sodium phosphate monobasic and dibasic, HPLC-grade water, acetonitrile (ACN), and methanol, which were of analytical grade.

Instrumentation

The instruments used for the quantification of CS, TF and Curcumin Tofacitinib Nanocarrier's (CS-TF-NCs) were a double Beam UV-

visible spectrophotometer (JASCO V-630), having wavelength scanner range from 800 nm to 190 nm used for UV-visible spectroscopy-related studies. A silica quartz square cell cuvette with dimensions of 3 cm in length and 1 cm in route length was used for the entire experiment, having a transmittance of 50.4±0.2%. For High-Performance Liquid Chromatography (HPLC) related studies, Agilet (1100) with Jasco UV 2075 Plus detector coupled with autosampler, a Jasco Lc-net 11/Adc valve, a G1310A Iso pump, a $\rm C_{18}$ (Agilent) id (4.6 x 250 mm) column and equipped with chemstation chromatography data system for analysis was used. An analytical balance (Shimadzu, Japan), a pH meter (Systronics, Ahmedabad, India), and other instruments were used for the study.

Preformulation study of curcumin and tofacitinib

The preformulation analysis and conformation of CS and TF were conducted to evaluate its purity by using various advanced methods, such as X-ray Diffraction (XRD), Differential Scanning Calorimeter (DSC). Fourier transform infrared spectroscopy (FTIR), followed by its organoleptic characteristics. The above studies were also conducted to assess the compatibility of curcumin and tofacitinib [13].

The compatibility between CS and TF was evaluated by X-ray Diffraction (XRD), Differential Scanning Calorimetry (DSC), and Fourier Transform Infrared Spectroscopy (FTIR). Combination-induced changes in crystallinity were measured using XRD, DSC examined possible thermal interactions through the detection of changes in melting points or exothermic/endothermic behavior, and FTIR determined probable chemical interactions by correlating functional group shifts in the spectra of the single drugs and the

combined drug. These tests assured the stability and compatibility of curcumin with tofacitinib prior to nanocarrier formulation [15, 16].

UV-visible spectroscopy-based quantification of pure curcumin, Tofacitinib, and a combination of Curcumin-tofacitinib and CTNs

Preparation of std. stock solution

A standard stock solution (1000 μ g/ml) of CS and TF was prepared by weighing 10 mg and adding in 10 ml methanol for curcumin and 10 ml water for tofacitinib, employing a 10 ml volumetric flask. The combination of Curcumin and Tofacitinib stock solution was prepared by weighing 5 mg each of curcumin and 10 ml methanol. This will be used for measurement of the maximum absorption of λ max of CS, TF, CS-TF, and CS-TF-NCs.

Preparation of a working solution

To achieve a concentration of 10 µg/ml, the working stock solution was further diluted with the respective solvent (1 ml to 10 ml). 1 ml of the standard stock solution was diluted to 10 ml to create the working stock solution (100 µg/ml).

Preparation of sample of CS-TF-NCs

A sample of CS-TF-NCs was prepared by measuring an accurate amount of CS-TF-NCs (equivalent to 5 mg each of CS and TF) and dissolving it in some amount of ethanol followed by transferring the resulting solution to a 50 ml volumetric flask and making up the volume to the mark by addition of ethanol to obtain a concentration of $100~\mu g/ml$. Further, 1 ml of the prepared solution was diluted in 10 ml of methanol to obtain a solution with a final concentration of $10~\mu g/ml$ (fig. 1).

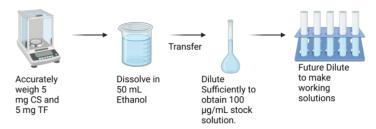


Fig. 1: Sample preparation of curcumin tofacitinib nanocarriers

Selection of \(\lambda \) max

The prepared solution was scanned against ethanol as a blank in the wavelength range of 200–400 nm. A prominent absorption peak was observed at 425 nm for curcumin and 286 nm for tofacitinib, which is depicted in the UV-visible spectra (fig. 2A). This wavelength (\$\lambdamax) was selected for further analysis. For the generation of the calibration curve, the stock solution was further diluted with ethanol to prepare a series of concentrations ranging from 2 to 10 µg/ml, which were then scanned at the respective wavelengths.

Validation of analytical method

The analytical method developed was validated in accordance with the ICH Q2 (R1) guidelines, which encompass aspects such as linearity, accuracy, precision, specificity, robustness, and LOD and LOQ. The linearity of the method was obtained through the analysis of serial dilutions from 1 to 180 $\mu g/ml$ and plotting the peak area against the concentration. Data obtained were confirmed for linearity through least-square regression analysis. The suitability of the method in terms of accuracy was further checked by spiking a 10 $\mu g/ml$ standard solution with excess amounts of the drug as 80 % (8 $\mu g/ml$), 100 % (10 $\mu g/ml$), and 120 % (12 $\mu g/ml$). Furthermore, SD and % RSD were calculated to assess the accuracy of the method [17].

Precision was established by preparation of three different concentration levels for the drug, that is, $5~\mu g/ml$, $10~\mu g/ml$, and $15~\mu g/ml$, at different time intervals on the same day (intra-day precision) and their repeated estimations on another day (inter-day precision). Then, mean percentage recovery, SD (Standard Deviation), and %RSD (Relative Standard Deviation) were calculated to establish the method's precision. The LOD and LOQ were obtained

from the slope of the linearity plot, S, and the standard deviation of the blank response, σ , and calculated by using equations 1 and 2.

LOD =
$$3.6\sigma/S$$
 Eq. 1
LOQ = $10\sigma/S$ Eq. 2

Analysis of six replicative analyses were used to assess the system's suitability on a standard solution of 30 $\mu g/ml$. Then, SD and %RSD for peak area, as well as retention time, were calculated. Finally, specificity was evidenced by concentrating a 30 $\mu g/ml$ sample solution prepared from the reference standard in conditions that might reasonably be expected to cause degradation and analysing any degradation products produced for their interference potential with the reference standard [18]

HPLC quantification of CS, TF, CS-TF and CS-TF-NCs

Chromatographic conditions

HPLC of CS, TF, CS-TF, and CS-TF-NCs was carried out using C_{16} [Agilent) column employing Methanol and Orthophosphoric acid pH 3.5 as mobile phase

Preparation of sample

Samples of CS, TF, CS-TF, and CS-TF-NCs were prepared using varying proportions of Methanol and water using HPLC, isocratic mode at a constant flow rate of 0.7 ml/min, the estimation was carried out by maintaining the column temperature 30°C the sample injection volume being $20~\mu\text{l**}$ for 10 min, and the detection is done at the respective wavelengths table 1. The mobile phase for RP-HPLC was optimized on the basis of both experimental trials and previous literature to obtain

maximum separation, peak resolution, and minimum tailing. Various ratios of methanol and aqueous orthophosphoric acid (pH 3.5) were tried, and a final ratio of 80:20 (v/v) methanol: orthophosphoric acid was chosen. This mixture gave sharp peaks, best retention times

(2.212 min for curcumin and 4.285 min for tofacitinib], and high sensitivity. The pH of 3.5 was selected because it improved peak symmetry and reduced peak broadening, which ensured high reproducibility in drug quantification [15].

Table 1: Chromatographic conditions for HPLC estimation of Curcumin, tofacitinib, and curcumin tofacitinib nanocarriers

S. No.	Sample	Column	Mobile phase	Flow rate	Detector
1	CS	$C_{18}[Agilent)$	Methanol: Orthophosphoric Acid pH 3.5 (80:20)	0.8 ml/min	Jasco UV 2075 Plus detector
2	TF	$C_{18}[Agilent)$	Methanol: Orthophosphoric Acid pH 3.5 (80:20)	0.8 ml/min	Jasco UV 2075 Plus detector
3	CS-TF-NCs	$C_{18}[Agilent)$	Methanol: Orthophosphoric Acid pH 3.5 (70:30)	0.8 ml/min	Jasco UV 2075 Plus detector

Method validation

The validation of High-Performance Liquid Chromatography (HPLC) for estimating CS and TF in the mobile phase was based on ICH guidelines Q2 (R1), including parameters such as linearity, precision, accuracy, repeatability, limit of detection (LOD), and limit of quantification (LOQ). The linearity was checked over a concentration range of 2-50 $\mu g/ml$ using a linear regression model and the least-squares method. A calibration curve was prepared by plotting the peak area against the analyte concentration of CS and TF in a mobile phase consisting of Methanol: water in an 80:20 ratio, using five standard solutions at concentrations of 2, 4, 6, 8, and 10 $\mu g/ml$ to calculate the regression line [16].

Precision and accuracy were determined through repeatability and intermediate precision studies. Repeatability was established by determining intraday variation, which consisted of three replicates at three levels of concentration (2, 6, and 8 $\mu g/ml)$ carried out at different times of the same day. Intermediate precision was checked by interday variation, where the same concentrations were measured over three different days. SD and RSD were used to check the accuracy of the data for intraday and interday variation. System suitability was further established by analyzing the CS and TF samples at a concentration of 10 $\mu g/ml$ in triplicate; tailing factor (T), height equivalent to a theoretical plate (HEPT), and the number of theoretical plates (N) were used to ascertain the system performance.

The robustness of the method was checked by minor changes in critical analytical parameters, such as variations in mobile phase composition (Methanol: Water at an 80:20 ratio), temperature fluctuations (25±1 °C), and flow rate alterations (0.7 ml/min and 1 ml/min). The test solutions were prepared at 2, 6, and 8 $\mu g/ml$ concentrations, and the data collected under the modified conditions were analyzed using RSD and percentage recovery to evaluate the performance of the method under the altered conditions.

The calculated values for LOD and LOQ are based on ICH guidelines, using the slope of the calibration curve (S) and the standard deviation for the response value. Subsequently, their results have been confirmed experimentally. A validated method can offer remarkably high reliability and good precision and hence proves robust enough to conduct quantification of CS and TF by its suitability in a prescribed mobile phase [17].

Statistical analysis

For every experiment, at least six experiments were carried out. Mean±standard deviation (SD) was used to present the data. Every piece of data was compared using one-way analysis of variance (ANOVA). A variety of techniques in addition to the Newman-Keul's test, a standardised test that corresponds with them. At p<0.05, a significant difference was determined [22].

RESULTS AND DISCUSSION

This research takes advantage of the synergistic action of curcumin (CS), a plant-based anti-inflammatory agent, and tofacitinib (TF), a Janus kinase inhibitor, on the IL-6/JAK/STAT3 pathway-a key pathway involved in breast cancer development and metastasis. Combination therapy approaches have been increasingly confirmed to have synergistic action in tumor inhibition and immune modulation. Consistent with earlier research, Bota *et al.* (2024)

demonstrated the therapeutic benefit of natural compound–drug combinations in lung cancer, affirming the synergy concept across cancer types [23].

In breast cancer, this two-mechanism strategy holds promise for increasing efficacy while minimizing drug resistance. The design of the current study mirrors the trend toward multi-targeted nanotherapy platforms, further affirming novel clinical approaches in precision oncology [24].

Preformulation studies of curcumin and tofacitinib

An analysis of organoleptic characteristics revealed that CS had a characteristic odour, yellow in color, and in the form of powder, while TF was odourless, white in colour, and an amorphous powder. Using a digital melting point apparatus (Electronics India 935, India), the melting point was observed to be within a range of 179-183 °C for CS and in range of 199-206 °C for TF, which corresponded with the reported reference (ELHAM et al., 2023). The melting point was also confirmed by, Differntial Scanning Calorimeter (DSC) (Mettler-Toledo, Japan). The range of determination was 20 to 250 °C. At 175.9 °C, the DSC thermogram (fig. 2A and 2D) displayed a strong peak and end set value indicating crystalline nature for CS and at 212.9 °C for TF. The FTIR spectra (fig. 2C and 2E) from Shimadzu, Japan, display the appropriate band at curcumin reveals key functional groups, including O-H (hydroxyl) stretching at 3700-3508 cm⁻¹, C=O (carbonyl) stretching at 1626-1599 cm⁻¹, and C=C (aromatic ring) vibrations around 1506 cm⁻¹. These confirm curcumin's aromatic, keto-enol structure with hydroxyl and alkyl groups [18].

DSC, XRD, and FTIR analyses confirmed the structural integrity and compatibility of CS and TF prior to formulation. No significant thermal or chemical interactions were observed, supporting the stability of the CS-TF mixture. The use of orthogonal techniques aligns with the standards described by Saatkamp *et al.* (2023), who recommended multi-modal preformulation analyses in drug development [15].

UV-Visible spectroscopy-based quantification of pure curcumin tofacitinib and CS-TF-NCs

It was observed that CS and TF were soluble in methanol. As indicated in fig. 2A and C, the absorption maximum (λ max) was determined to be 426 nm and 286 nm, respectively. With a correlation value $(R^2) = 0.9962$ (fig. 2B) and $(R^2) = 0.998$ (fig. 2D) respectively high linearity within the concentration range of 10-50 μg/ml table 2, the regression equation of the curve was determined to bey = 0.0317x+0.0114 at 426 nm and y = 0.026x+0.121 at 286 nm, respectively. The absorption maximum (λ max) for CS-TF-NCs was determined to be 287 nm and 421 nm as indicated in (fig. 3 A) having a high linearity in the concentration range of 5-25 µg/ml table 2 with a correlation value (R^2) = 0.999 at 287 nm and (R^2) = 0.9994 at 421 nm, the regression equation of the curve was determined to be y = 0.0186x+0.0072 at 287 nm and y = 0.0117x-0.0081 (fig. 3B). The precision intra-day and inter-day data shows strong reproducibility with a percentage RSD less than 1.5% in table 3, indicating that the approach is precise. It was discovered that the mean recovery value (table 4) at various doses was greater than 95%, the procedure is accurate. CS, TF and CS-TF-NCs LOD and LOQ were discovered to be 0.012 µg/ml, 0.027µg/ml and 0.04 µg/ml, respectively and were reported in table 5 [19, 27].

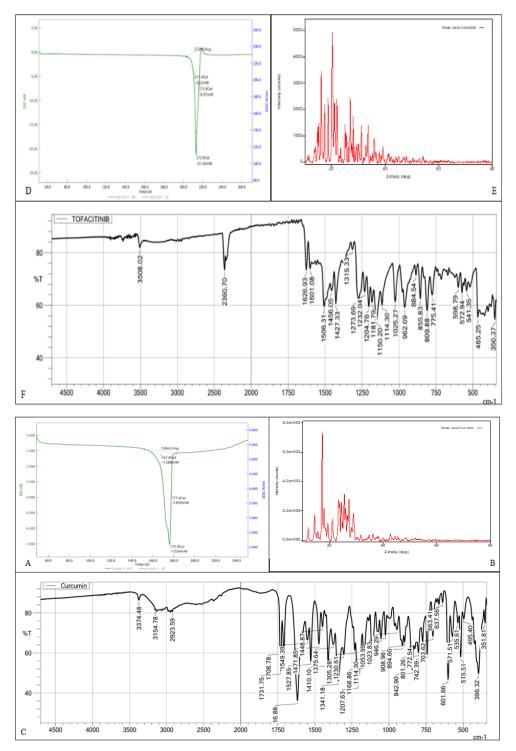


Fig. 2: Preformulation study of curcumin (A) DSC, (B) XRD, (C) FTIR and Tofacitinib = (D) DSC, (E) XRD, (F) FTIR

Table 2: Data for calibration curve of curcumin, tofacitinib, CS-TF, and CS-TF-NCs

S. No.	Concentration (µG/ml)	Absorbance			
		CS	CS TF CS-TF-NCS		
				AT 287 NM	AT 427 NM
1	10	0.376±0.005	0.317±0.003	0.099±0.002	0.053±0.006
2	20	0.665±0.007	0.664±0.008	0.192±0.006	0.106±0.007
3	30	0.896±0.003	0.982±0.001	0.292±0.004	0.165±0.006
4	40	1.197±0.004	1.23±0.01	0.372±0.005	0.225±0.001
5	50	1.43±0.01	1.619±0.009	0.473±0.008	0.285±0.004

All the values are expressed as mean±SD, n=6

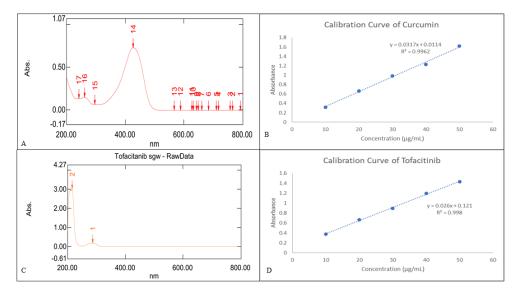


Fig. 2: (A) λmax of Curcumin, (B) Calibration curve of Curcuminat 426 nm, (C)λmax of Tofactinib, (D) Calibration curve of Tofactinib 277 at nm

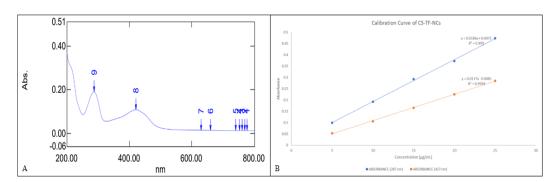


Fig. 3: (A) \(\lambda\) max of CS-TF-NCs, (B) Calibration curve of CS-TF-NCs

Table 3: Results of intra- and inter-day precision

Concentratio	Intra day						Inter day					
n (μG/ml)	CS		TF	TF CS-T		CS-TF-NCS CS			TF		CS-TF-NCS	
	Absorbanc	%RSD	Absorbanc	%RS	Absorbanc	%RS	Absorbanc	%RS	Absorbanc	%RS	Absorbanc	%RS
	e (NM)±SD		e (NM)±SD	D	e (NM)±SD	D	e (NM)±SD	D	e (NM)±SD	D	e (NM)±SD	D
20	0.661±0.04	0.0915	0.666±0.07	0.093	0.192±0.006	0.09	0.68±0.02	0.92	0.67±0.05	0.097	0.188±0.003	0.07
20	0.670±0.02	0.0927	0.661±0.05	0.097	0.187±0.004	0.08	0.678±0.05	0.97	0.668±0.06	0.095	0.182±0.007	0.09
20	0.660±0.05	0.097	0.664±0.08	0.091	0.191±0.002	0.05	0.675±0.05	0.95	0.670±0.05	0.093	0.191±0.007	0.05

All the values are expressed as mean±SD, n=6

Table 4: Recovery study of CS, TF and CS-TF-NCs

λmax (NM)	Level of recovery (%)	Amount spiked recovery (mg/ml)	Amount recovered (mg/ml)	Recovery (%)	Mean recovery
CS					
426 nm	140	14	12.89±0.28	92.1%±0.2%	
	160	16	15.31±0.21	95.6%±0.4%	94.4 %±0.5%
	180	18	17.21±0.18	95.6%±0.7%	
TF					
277 nm	140	14	13.21±0.23	94.3%±0.6%	
	160	16	15.43±0.31	96.4%±0.3%	95.5%±0.6%
	180	18	17.25±0.25	95.8%±0.5%	
CS-TF-NCs					
287 and 421	150	15	14.7±0.32	98%±0.9%	
nm	180	18	17.8±0.21	98.8%±0.1%	98.1%±0.3%
	200	20	19.5±0.18	97.5%±0.8%	

All the values are expressed as mean \pm SD, n=6

Table 5: Validation parameters

Validation parameters	Results		
_	CS	TF	CS-TF-NCS
Λmax	426 nm	277 nm	287 and 421 nm
Beer's law range (μg/ml)	10-50 μg/ml	10-50 μg/ml	5-25 μg/ml
Correlation coefficient (R2)	0.9962	0.998	0.999 and 0.9994
Slope (m)	0.317	0.026	0.0186 and 0.0117
Intercept (c)	0.0114	0.121	0.0072 and 0.0081
Accuracy	94.09	95.12	97.15
Precision (%RSD)	0.309	0.255	0.215
Intra-day	0.0996	0.2064	0.095
Inter-day	0.1051	0.1029	0.099
LOD (μg/ml)	0.012	0.027	0.045
LOQ (μg/ml)	0.03	0.055	0.07

RSD: Relative standard deviation, LOD: Limit of detection, LOQ: Limit of quantification.

Compared to Ahmed *et al.* (2015), who validated a UV method for tolfenamic acid in simple matrices, this study extends UV applicability to a complex nanocarrier system, offering higher sensitivity and broader pharmaceutical utility [28].

UV-spectrophotometric analysis revealed distinct λ max values for CS (426 nm) and TF (286 nm), which shifted to 421 and 287 nm upon nanocarrier encapsulation. This shift may indicate molecular interactions or alterations in the microenvironment around the drugs. The high correlation coefficient (R² = 0.9994) in the range of 5–25 µg/ml demonstrates excellent linearity, while the LOD and LOQ (0.012 µg/ml and 0.03 µg/ml for CS) confirmed high sensitivity.

High-performance liquid chromatography

To estimate CS and TF in formulated nanocarrier, a validated HPLC method was developed. The HPLC approach is superior to UV spectrophotometric assessment because it allows for approximation even at lower concentrations. In addition, this technique's simple actions make it easier to detect even at very small concentrations Depending on the solubility of CS and TF, different solvents such as ethanol, methanol, and 2-propanolol were tried, and their proportions were adjusted with water. This was the first step in the isocratic chromatographic quantification of CS and TF. Numerous trial runs were conducted with different combinations of methanol and orthophosphoric acid pH 3.5, and the chromatograms produced showed uneven peaks, poor separation, and tailing factors less than 5. Simultaneously, the tailing factor was less than 3 and the ratio Methanol: orthophosphoric acid pH 3.5 (75:25) provided

comparatively acceptable resolution. The best separation of CS and TF was obtained using Methanol: orthophosphoric acid pH 3.5 (80:20), with chromatographic parameters including a column temperature of 26 °C, an injection volume of 20 μl, a flow rate of 0.8 ml/min, and a detection wavelength of 302 nm [29]. Under ambient settings, 2.630±0.005 and 2.631±0.003 was the retention time (Rt) at which CS and TF were discovered respectively (fig. 4A). The theoretical plate (TP) value of 2268±55 and 2268±67 and the factor (TF) of 0.91±0.003 and 0.89±0.007 were both within the predetermined bounds table 6. By maintaining the other chromatographic conditions as previously described, the primary goal of analytical profiling of CS, TF, and CS-TF-NCs was achieved. The solubility of CS-TF-NCs was checked by using different solvents such as ethanol, methanol, isopropyl alcohol and other polar solvents, CS-TF-NCs was found to be soluble in methanol and its proportion was adjusted with orthophosphoric acid pH 3.5, isocratic mode of HPLC was used for quantification of nanocarrier which showed uneven peaks and tailing factors less than 3. Simultaneously, the tailing factor was less than 3.5 and the ratio Methanol: orthophosphoric acid pH 3.5 (60:40) provided comparatively acceptable resolution. The best separation of CS-TF-NCs was obtained using Methanol: orthophosphoric acid pH 3.5 in a ratio of 80:20, with chromatographic parameters including a column temperature of 26 °C, an injection volume of 20 µl, a flow rate of 0.8 ml/min, and a detection wavelength of 302 nm. Under ambient settings, 2.212±0.008 and 4.285±0.01was the retention time (Rt) at which CS and TF were discovered, respectively (fig. 4B). The theoretical plate (TP) value of 3968±45 and 6668±77 and the factor (TF) of 0.85±0.005 and 0.75±0.003 table 6.

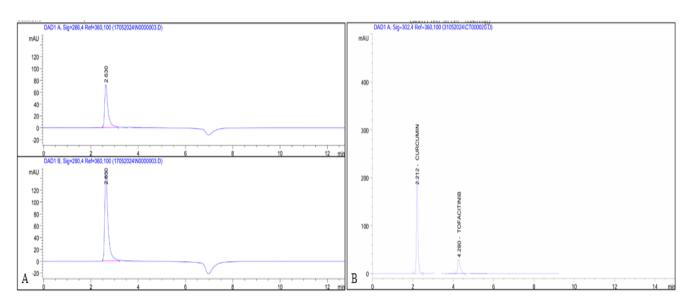


Fig. 4: HPLC chromatograms of (A) Curcumin and Tofactinib in mobile phase (B) Curcumin-Tofacitinib-Nanocarriers. Chromatographic conditions were injection volume 20 µl**; flow rate 0.8 ml/min; temperature 25 ° C; run time 10 min; detection wavelength 302 nm

Table 6: System suitability parameters

Parameter	CS		TF		CS-TF-NCS		
	Mean±SD	%RSD	Mean±SD	%RSD	Mean±SD	%RSD	
Rt	2.630±0.005	0.172	2.631±0.003	0.165	2.212±0.008 and 4.285±0.01	0.25	
TF	0.91±0.003	0.1	0.89±0.007	0.15	0.85±0.005 and 0.75±0.003	0.13	
TP	2268±55	2.7	2268±67	3.1	3968±45 and 6668±77	2.1	

All the values are expressed as mean±SD, n=6

The RP-HPLC technique provided better resolution and detection. Well-defined peaks with retention times of 2.212 min (CS) and 4.285 min (TF) and low LODs (0.045 μ g/ml) confirm the sensitivity of the system. Recovery of 97% to 99%, and RSD values always<1.5%, confirm precision and repeatability.

Relative to Gorantla *et al.* (2021), who prepared HPLC protocols for TF in liquid crystalline nanoparticles, this work breaks new ground through the validation of the simultaneous detection of both drugs within a combination nanocarrier system [13]. In addition, Trivedi *et al.* 2021) provided HPLC-based quantitation of thymoquinone with greater LODs, implying that the current method is superior to earlier models in terms of sensitivity [18].

Method validation

The calibration curve was used to verify the linearity of the devised analytical method by examining the relationship between the response and the concentration of CS and TF in the

sample. A calibration curve was built (n = 6) for the CS and TF in the mobile phase Methanol: Orthophosphoric acid pH 3.5 in a ratio of 80:20 for five concentrations ranging from 5 to 25 $\mu g/ml$. The regression equations for the Methanol: Orthophosphoric acid pH 3.5 in a ratio of 80:20, respectively, were found to be y = 0.0257x+0.0102; correlation coefficient (R²) = 0.9951 and it was determined from the equations and quality of curve that the results were significant and linear, with a rise in response observed as the concentration of CS and TF in the sample increased [30, 31].

The study examined the repeatability and intermediate precision of CS and TF at three distinct doses. In the mobile phase, the concentrations were 5 $\mu g/ml$ (low), 10 $\mu g/ml$ (medium), and 15 $\mu g/ml$ (high) table 7. The analysis was conducted on the same day (repeatability) and over three separate days (intermediate precision). The maximum %RSD values table 8 were less than 2, indicating that the developed approach had higher precision.

Table 7: Recovery study

Concentration (µG/ml)	CS	TF	CS-TF-NCS	
	% Recovery	% Recovery	% Recovery	
5	97.7±0.29%	97.6±0.27%	98.6±0.09%	
10	97.8±0.25%	98.9±0.11%	98.8±0.06%	
15	98.5±0.23%	99.7±0.33%	98.9±0.07%	

All the values are expressed as mean±SD, n=6

Table 8: Repeatability and intermediate precision

Concentration	Inter day						Intra day					
(μg/ml)	CS TF		TF		CS-TF-NCS		CS TF		TF	CS-TF-NCS		
	Mean±SD	%RSD	Mean±SD	Mean±SD	%RSD	Mean±SD	Mean±SD	%RSD	Mean±SD		Mean±SD	%RSD
5	99.21±0.07	0.35	99.11±0.2	0.317	98.86±0.23	0.185	97.83±0.39	0.317	99.31±0.09	0.215	99.03±0.12	0.317
10	98.2±09	0.28	95.07±03	0.279	98.75±31	0.165	96.21±13	0.279	98.91±10	0.186	97.87±07	0.279
15	99.87±0.27	0.34	99.17±0.4	0.208	98.01±0.05	0.238	98.05±0.08	0.208	99.10±0.32	0.149	98.90±0.27	0.208

All the values are expressed as mean±SD, n=6

The standard deviation of the response and the slope obtained from linear regression of the calibration curve were used to calculate the LOD and LOQ within acceptable precision and accuracy. The minimum quantity of CS, TF and CS-TF-NCs that can be found, or LOD and LOQ in Methanol: Orthophosphoric acid pH 3.5 in a ratio of 80:20, For n=6, the corresponding values were 27.95 ± 0.67 ng/ml, 31.11 ± 0.45 ng/ml, and 97.09 ± 0.5 ng/ml, 99.51 ± 0.37 ng/ml.

A comparative analysis of UV-spectrophotometry and RP-HPLC for curcumin-tofacitinib nanocarriers showed high precision, sensitivity, and accuracy. UV-spectrophotometry had a linearity range of $10\text{--}50~\mu\text{g/ml}$ for CS and TF with excellent correlation coefficients (R 2 = 0.9962 and R 2 = 0.998). The LOD values were 0.012 $\mu\text{g/ml}$ (CS) and 0.027 $\mu\text{g/ml}$ (TF), and LOQ values were 0.03 $\mu\text{g/ml}$ (CS) and 0.055 $\mu\text{g/ml}$ (TF), indicating that the method was of high sensitivity.

RP-HPLC analysis yielded retention times of 2.212 min (CS) and 4.285 min (TF) (fig. 4). The approach was found to be highly precise, with %RSD<1.5% at all concentrations tested table 8. Recovery levels varied between 97% and 99%, proving the accuracy and reproducibility of nanocarrier quantitation.

The established methods present a solid analytical protocol for pharmaceutical uses, assuring accurate drug measurement, stability testing, and quality control in nanocarrier formulations.

CONCLUSION

This research effectively formulated and validated sophisticated analytical techniques for the quantification and characterization of curcumin-tofacitinib nanocarriers (CS-TF-NCs) to overcome solubility, bioavailability, and systemic toxicity limitations. The improved UV-spectrophotometry and RP-HPLC techniques were highly sensitive, accurate, and reproducible with Amax values of 287 nm and 421 nm and $\rm R^2=0.9994$ in a 5–25 $\mu g/ml$ range. Recovery yields for CS, TF, and CS-TF-NCs were 94.09%, 95.12%, and 97.15%, respectively, with retention times of 2.212 and 4.285 min under the optimal chromatographic conditions. LOD and LOQ were 0.045 $\mu g/ml$ and 0.07 $\mu g/ml$, validating the precision of the method for use in pharmaceutical applications.

Aside from these observations, further studies are needed to generalize this method's use in other combination therapies, explore its use in different drug delivery systems, and perform in vivo studies to confirm therapeutic effectiveness in breast cancer models. These follow-up studies will continue to enhance the clinical significance and translational value of curcumin-tofacitinib nanocarriers in oncology.

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AUTHORS CONTRIBUTIONS

Experimentation and data analysis were done by Ms. Suchita Waghmare, Manuscript drafting and experimentation work was carried out by Mr. Ujban Hussain, the concept for the work was proposed by Nilesh Rarokar, this work was carried out under the supervision of Dr. P. B. Khedekar.

CONFLICT OF INTERESTS

The writers confirm that their work was not impacted by personal or financial interests.

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