

UNLOCKING THE POTENTIAL OF BETALAIN COMPOUNDS FROM BEETROOT: TARGETING *STREPTOCOCCUS MUTANS* IN DENTAL CARIES THROUGH MOLECULAR DOCKING, ADMET PROFILING, AND ANTIBACTERIAL ACTIVITY

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ABSTRACT

Objective: Dental caries, primarily caused by *Streptococcus mutans*, pose a significant global health challenge, often treated with fluoride and synthetic dyes like erythrosine, despite associated toxicity risks. This study explores betalains, natural pigments from *Beta vulgaris* (beetroot), as dual-function agents for plaque detection and biofilm inhibition.

Methods: Ten betalains were evaluated through *in silico* absorption, distribution, metabolism, excretion, and toxicity (ADMET) analysis and molecular docking against glucosyltransferase and antigen I/II (Ag I/II), key proteins in biofilm formation. Antibacterial activity against *S. mutans* was performed to determine the inhibition zone and minimum inhibitory concentration (MIC) of beetroot extracts.

Results: Docking validation showed root mean square deviation (RMSD) values below 4 Å, confirming reliability. Isobetanin exhibited the strongest binding affinities (-10.427 and -10.893 kcal/mol) and interacted with active residues GLU515 and ASP477, crucial for biofilm formation. High solubility, low toxicity, and limited systemic absorption make betalains ideal for topical applications, such as dental disclosing solutions. *In vitro* studies have shown that the beetroot ethanol extract from Magelang has higher antibacterial activity than betalains. The beetroot ethanol extract contains not only betalains but also other compounds that synergistically inhibit the growth of *S. mutans*.

Conclusion: These findings highlight betalains as safer, natural alternatives to synthetic dyes, paving the way for innovative and sustainable dental care formulations with enhanced safety and efficacy. Additionally, further study needs to determine the effectiveness of the extracts in inhibiting dental caries formation.

Keywords: Betalains, *Beta vulgaris*, *Streptococcus mutans*, Biofilm inhibition, Glucosyltransferase, Antigen I/II, Dental disclosing solution

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INTRODUCTION

Streptococcus mutans is a bacterium named as the main cause of dental caries. The microorganism is strongly associated with plaque formation and acid production, which contributes to further enamel deterioration and increases the risk of tooth decay [1]. Dental caries, commonly known as tooth decay, is a pervasive global health issue affecting billions of individuals across all age groups. According to the World Health Organization (WHO), approximately 2 billion people suffer from caries in permanent teeth, while caries in primary teeth impact 514 million children [2]. Current treatments for dental caries primarily rely on fluoride-based methods to inhibit bacterial growth and strengthen enamel. While effective, these approaches have notable limitations. Detection kits for tooth decay, often employing fluoride-based solutions and erythrosine-based plaque-disclosing agents, also face significant challenges. Fluoride, though effective, poses risks of dental fluorosis and potential neurotoxic effects, sparking debates over its safety. Similarly, erythrosine, a synthetic dye used in plaque detection, faces scrutiny due to cytotoxic concerns, leading to restrictions in some countries [1, 3]. These limitations have driven interest in natural alternatives that showed minimum side effects and had broad antimicrobial potential, like betalains and bioactive pigments from beetroots with antioxidant and antimicrobial properties [4, 5]. Betalains are potential dual-action solutions for plaque detection and bacterial inhibition, offering a safer and more sustainable approach to advancing oral healthcare globally [6].

Several recent works on betalains as colored pigments of various plants have pointed out their health-beneficial potential, especially concerning antioxidant and antimicrobial action. Its rich sources can be extracted from *Beta vulgaris*, commonly known as beetroot [7, 8].

These molecules have attracted great interest for their applications in promising biological and therapeutic contexts, such as fighting oxidative stress and bacterial infections. Despite their huge potential, betalains' effectiveness in inhibiting the activity of *S. mutans*, which causes dental caries, is unrevealed in many health-related fields [9]. Therefore, further investigation is needed into the potential role of betalains as an alternative supplement therapy targeting *S. mutans* for better oral health. Alternatively, instead of this, betalain and other constituents from beetroot could be developed as a detection kit for dental caries.

Beetroot comprises mainly betacyanins and betaxanthins. Major betacyanins are betanin, isobetanin, neobetanin, 15-decarboxybetanidin, and 17-decarboxyisobetanidin, whereas vulgaxanthin I, vulgaxanthin IV, valine-betaxanthin, γ -Aminobutyric acid-betaxanthin, and isoleucine-betaxanthin are some major betaxanthins [10]. All of them have tremendous potential to be developed into a caries detector, mainly in the form of a disclosing solution. With their colorful appearance and compatibility with living tissues, betalains may become a substitute for natural, nontoxic synthetic dyes in dental plaque detectors. Their inherent bioactive properties, entailing antioxidant and antimicrobial activity, would suggest that the disclosing solution made from betalain may not only exert an inhibitory effect on the formation of dental caries but also have a therapeutic effect by acting on disease-causing oral bacteria, including *S. mutans* [11, 12]. This dual functionality presents an innovative and scientifically sound approach to enhancing dental diagnostics and preventive care.

Natural pigments, such as anthocyanins, have been considered for therapeutic and diagnostic oral applications; however, their use is

discouraged due to their poor stability, as they are highly susceptible to degradation by pH variation, light, oxidation, and elevated temperatures [1]. By comparison, betalains exhibit greater stability across a wider pH range and greater resistance to thermal and photodegradation [2]. For instance, it was recently demonstrated that under storage in the dark, betalain pigments extracted from *B. vulgaris* exhibited appreciable half-lives, and exposure to light increased degradation substantially [3]. Additionally, comparative overviews of natural pigment stability report that betalains are more robust to a greater range of intrinsic and extrinsic stressors than anthocyanins [3, 4]. Taken together, these findings strengthen the position of betalains as superior candidates to anthocyanins for dual-functional as detection and inhibition agents for dental plaque applications.

Glucosyltransferases are the sucrose-hydrolyzing enzymes produced by *S. mutans*, which catalyze the formation of glucans elaboration from sucrose that forms the extracellular polysaccharide matrix, playing an essential role in the biofilm formation and bacterial adherence on tooth surfaces. Glucans strengthen the biofilm and promote bacterial colonization and acid production, which, in turn, cause enamel demineralization and dental caries. Hence, glucosyltransferase was a key target for anti-caries therapies aimed at disrupting glucan synthesis [13]. Besides glucosyltransferase, antigen I/II (Ag I/II) is a surface adhesin protein of *S. mutans*, promoting bacterial attachment to salivary glycoproteins and tooth surfaces, thus playing a crucial role in biofilm formation. Ag I/II facilitates bacterial adherence and interaction with host components [14]. Therefore, Ag I/II has an important role in developing dental plaque and caries, making this antigen another important target for therapeutic intervention.

An efficient but powerful strategy chosen in this research is the *in silico* approach, which enables the screening and prediction of interaction between natural compounds and bacterial targets, facilitating the precise identification of therapeutic agents. These computational tools assess several compounds without requiring many time-consuming and costly laboratory experiments. Molecular docking enables observation in some instances to make certain predictions on the disposition of bioactive compounds, including betalains, to bind specific bacterial proteins. Its study may predict the binding affinity of active constituents to surface proteins involved in the biofilm formation of *S. mutans*, which has an important role in bacterial adherence and virulence. *In silico* methods through molecular docking and absorption, distribution, metabolism, excretion, and toxicity (ADMET) analysis will provide provisional data for the identification of putative anti-biofilm agents and thus provide better efficiency in drug discovery [15–18]. This would be one useful approach toward reducing time and resources in developing new dental diagnostics and preventive care. Researchers also conducted *in vitro* tests to determine the effectiveness of beetroot ethanol extract and betalain in inhibiting the growth of *S. mutans*. Therefore, this study aimed to conduct *in silico* and *in vitro* tests to explore the potential of *B. vulgaris*, commonly known as beetroot, to develop a new diagnostic tool for diagnosing and preventing dental caries.

MATERIALS AND METHODS

Plant materials

B. vulgaris L. acquired were collected from the local farmer in three different locations, including West Bandung, Ciwidey, and Magelang. The plants have been identified at Indonesian Herbarium Bogoriense, Directorate of Scientific Collection Management, National Research and Innovation Agency Republic (BRIN).

Extraction of samples

The fresh beetroots were cleaned in tap water and peeled, then sliced into thin portions. Each sample then was dried in oven at 40 °C for 48 h with Flap 50%. The dried beetroots then milled into powder. Each powder of beetroots from 3 different locations (100 g) was extracted with 70% ethanol in ratio 10 ml/g at room temperature using the Ultrasonic Assisted Extraction (UAE) for 2 h with three repetitions, respectively. The UAE was conducted using an ultrasonic bath (WT-600-40, Japan) with indirect contact and sonication applied in continuous mode with a power of 600 W and a

frequency of 40 kHz. Subsequently, each filtrate underwent evaporation using a rotary evaporator and was then freeze-dried to yield dried extracts, Bandung Barat (49.96%), Ciwidey (43.38%), and Magelang (27.46%).

Determination of the betanin content

In this study, the quantification of betanin content was conducted using Waters' ACQUITY UPLC System. Calibration curves were established from standard stock solutions of betanin, following the procedure reported by Rotich *et al.* (2022), with minor modifications [19]. The calibration was performed using betanin solutions at concentrations of 150, 200, 250, 300, 350, and 400 ppm prepared in MeOH/H₂O (1:1). Prior to analysis, samples were filtered through a 0.22 µm nylon syringe filter into Waters amber vials. The UPLC system was equipped with a PDA detector set at 534 nm and an ACQUITY UPLC BEH C18 column (130 Å, 1.7 µm, 2.1 × 50 mm). Analyses were carried out in triplicate under the following conditions: isocratic elution with acetonitrile (50%) and ammonium formate buffer (0.7 mmol) in water (50%) as the mobile phase, a flow rate of 0.4 ml/min, and an injection volume of 4 µL. The column temperature was maintained at 25 °C. Data acquisition and processing were performed using Empower 3 software.

Molecular docking

Data mining

A series of ligands compromised betalains, the major active constituents of *B. vulgaris*, were selected ten candidates for inhibiting *S. mutans* including vulgaxanthin-I (Chemspider ID: 32702057), isobetanin (Chemspider ID: 4884836), neobetanin (Chemspider ID: 103883522), valine-betaxanthin (PubChem CID: 157010371), γ-aminobutyric acid-betaxanthin (PubChem CID: 136747659), betanin (PubChem CID: 6540685), 17-decarboxyisobetanidin (CAS RN: 104331-28-4), 15-decarboxybetanidin [10, 20], isoleucin-betaxanthin and vulgaxanthin IV [10, 21]. All the structures are depicted in fig. 1. Those ligands were docked with glucosyltransferase and antigen I/II (Ag I/II) from *S. mutans*, both which play crucial roles in biofilm formation in tooth decay [14].

Drug-likeness and pharmacokinetic analysis

The Drug-likeness and pharmacokinetics properties analysis of betalains were evaluated using the SwissADME server (<http://www.swissadme.ch/>) based on Lipinski, Ghose, Veber, Egan, and Muegge. Additionally, toxicological assessments, including mutagenicity, carcinogenicity, hepatotoxicity, estrogenicity, androgenicity, acute oral toxicity, skin irritation, and eye irritation, were conducted using the VenomPred server (<https://www.mmvs.it/wp/venompred/>).

Ligands' preparations

The ligands consist of ten active constituents from *B. vulgaris* (fig. 1). Ligand structures were constructed using Avogadro [22]. The ligand's energy was minimized, and the geometry structure was optimized using the UFF force field by using 4 steps per update and steepest descent algorithm. The ligands were converted to .pdbqt format by using PyRx 0.8 [21]. Protonation state of ligand is not adjusted.

Receptor preparations

The crystal structures of glucosyltransferase [Protein Data Bank (PDB) ID: 3AIC] and Ag I/II (PDB ID: 3IPK) from *S. mutans* involved in biofilm growth were downloaded from RCSB Protein Data Bank (PDB): (<https://www.rcsb.org/>). Water molecules and ligands present in the protein structure were removed using UCSF Chimera, and the apo structure of the protein was saved in .pdb format. Then the protein's structure was converted to .pdbqt format by using PyRx 0.8 [21].

Docking protocol validation

Validating the docking methods involved docking the native ligands to the apo structure of the protein using PyRx 0.8 with Vina wizard [23, 24]. Then, the structure of the re-docked native ligand and native ligand on crystal structure were compared. Root mean square deviation (RMSD) was determined by using DockRMSD server

<https://zhanggroup.org/DockRMSD/> [25]. RMSD value of less than 2Å is excellent and RMSD value of less than 4Å is acceptable, suggesting that the docking protocol used in this study has been carried out successfully [26, 27].

Molecular docking between betalains and surface protein of *Streptococcus mutans*

Docking was conducted using PyRx 0.8 [23]. Previously optimized ligands and receptors were selected and docked with the Vina wizard feature by using the Run Vina option [24]. The grid box was set as follows to cover the known active site and native ligand on crystal structure. 3AIC (center_x = 190.3478, center_y = 45.6187, center_z = 199.5867, size_x = 30.5268, size_y = 22.8338, size_z = 22.0223). 3IPK (center_x = 6.7079, center_y = 44.8050, center_z = 23.1114, size_x = 34.1401, size_y = 32.2391, size_z = 33.8503).

Analysis and visualization

The best docking results were selected and saved in .pdb format by using UCSF Chimera [28]. Then the protein structure and the docked ligand were combined by using Pymol (<https://www.pymol.org/support.html?>). Ligand-protein interaction was visualized using PyMol and Discovery Studio software. The active sites of receptor proteins in this study were predicted by PrankWeb (<https://prankweb.cz/>) with a preference for custom structure used in receptor preparation.

Disk diffusion assay

Inhibitory activity was carried out using the disc method with slight modifications [29]. *S. mutans* was incubated with brain heart infusion (BHI) agar for 24 h at 37 °C. The inoculum was prepared by mixing several bacterial colonies with a sterile ringer solution until a bacterial suspension of 1×10^6 CFU/ml was obtained. The inoculum was taken and streaked on the surface of BHI agar. The samples, betanin, and amoxicillin were dissolved in BHI media. Sterile paper discs with a diameter of 6 mm that had been impregnated with 20 μ l* (1000 mg/ml) of the sample were placed on BHI agar. Betanin (1000 mg/ml) and Amoxicillin (0.5 mg/ml) were used as positive controls. The discs were then incubated for 24 h at 37 °C. The diameter of the inhibition zone (mm) was measured from the sterile paper disc and the clear zone formed.

Determination of minimum inhibitory concentration

Microdilution test with slight modification was conducted to determine the minimum inhibitory concentration of an extract on bacterial growth [29]. A bacterial suspension of 1×10^8 CFU/ml (0.5 McFarland) was prepared, diluted with physiological NaCl solution to obtain a bacterial suspension of 1×10^6 CFU/ml and then added with a two-fold dilution of the sample into the bacterial suspension in a 96 well plate, incubated for 24 h at 37 °C. The MIC value is the lowest concentration that can inhibit bacterial growth.

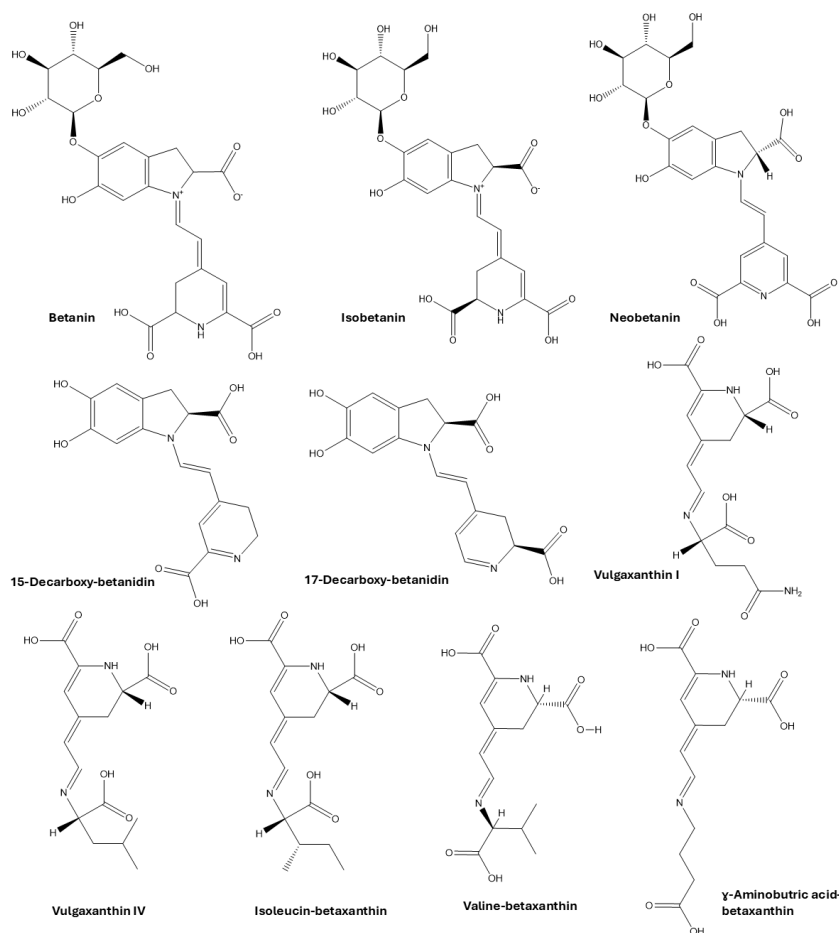


Fig. 1: The chemical structures of betalains from *Beta vulgaris* used in this study

RESULTS AND DISCUSSION

Quantification of betanin content

The UPLC chromatograms of betanin standards at concentrations of 150, 200, 250, 300, 350, and 400 ppm are shown in fig. 2 (a-f). The

betanin was consistently detected at a retention time (Rt) of 0.44 min, with a maximum absorption wavelength in the range of 534-538 nm. These chromatographic profiles were used to construct the calibration curve (fig. 3), which enabled the quantification of betanin in beetroot extracts.

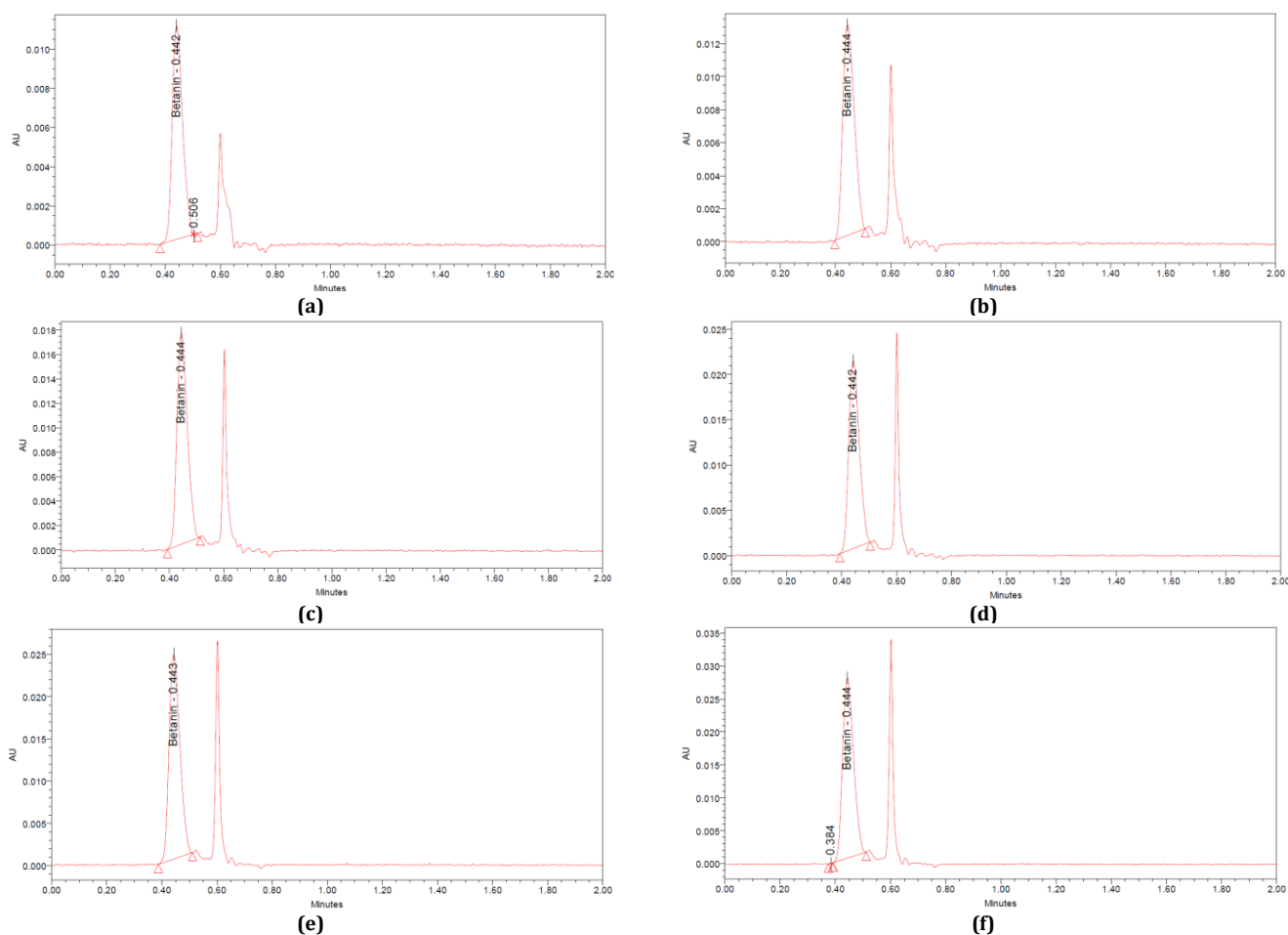


Fig. 2: Chromatograms for the betanin standard at 150 ppm (a), 200 ppm (b), 250 ppm (c), 300 ppm (d), 350 ppm (e), and 400 ppm (f)

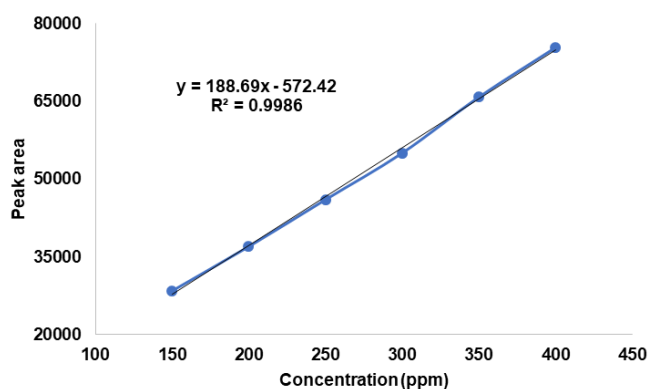


Fig. 3: UPLC calibration curve for betanin standards

The UPLC chromatograms of beetroot extracts are presented in fig. 4 (a-c). Each extract displayed a distinct chromatographic profile, although the characteristic betanin peak was consistently observed at Rt 0.443– 0.447 min. Quantitative analysis using the calibration curve revealed marked differences in betanin content among the samples (table 1). The amount of betanin present in the

beetroot was expressed in mg betanin/g dry extract of beetroot. Sample Ciwidey contained the highest concentration of betanin (845.95 mg/g), followed by sample Bandung Barat (748.12±2.72 mg/g). By contrast, sample Magelang exhibited only 15.62±0.24 mg/g, representing the lowest betanin level among the extracts analyzed.

Table 1: Betanin concentration of BB, C, and M beetroot extracts

No	Samples	Retention time	Peak area	Quantity estimates (mg/g)
1	Bandung Barat	0.443	27660±102.63	748.12±2.72
2	Ciwidey	0.446	31351±39.26	845.95±1.04
3	Magelang	0.447	17±9.09	15.62±0.24

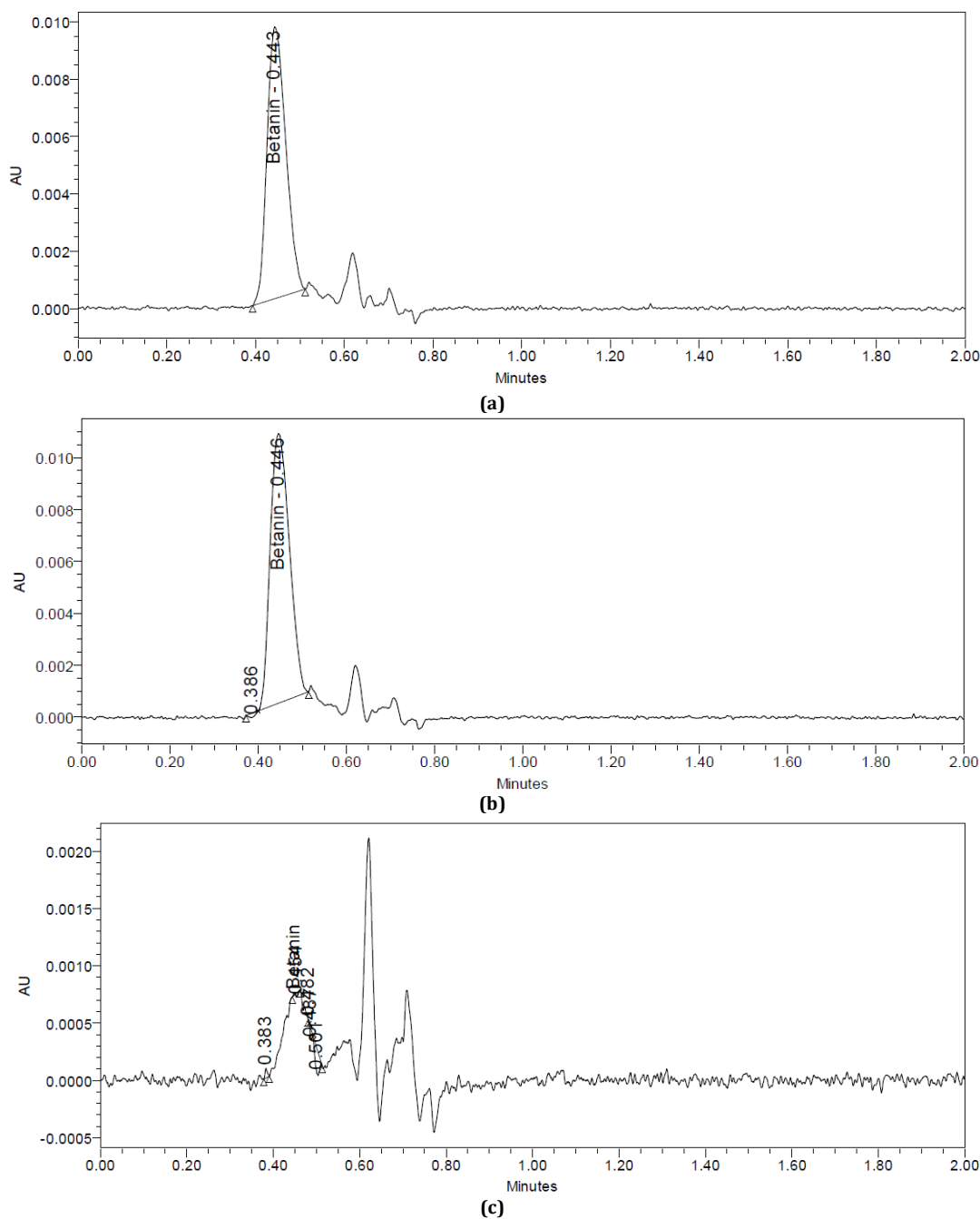


Fig. 4: UPLC chromatogram dried beetroot extract of Bandung Barat (a), Ciwidey (b), and Magelang (c)

Despite its relatively low betanin concentration, sample Magelang demonstrated additional chromatographic peaks at longer retention times, suggesting a higher abundance of other phytochemical constituents compared to samples Bandung Barat and Ciwidey. Based on the reports by Kuszniereicz *et al.*, 2021, the longer retention times in beetroot extract may be attributed to the betalamic acid and betaxanthin compounds [10]. In addition, the notably higher betanin levels in samples Bandung Barat and Ciwidey suggest that these extracts may possess stronger betacyanin pigment-related bioactivities, whereas sample Magelang may be richer in other phenolic or betaxanthin-pigmented compounds contributing to its overall phytochemical profile [30].

Drug-likeness and pharmacokinetics study

Betalains, pigment compounds isolated from *B. vulgaris*, are two principal groups: betacyanins and betaxanthins. The main constituents of betacyanins are betanin, isobetainin, neobetainin, 15-

decarboxybetanidin, and 17-decarboxy isobetainin, while representative betaxanthins are vulgaxanthin I, leucine betaxanthin/vulgaxanthin IV, valine betaxanthin, γ -aminobutyric acid betaxanthin, and isoleucine betaxanthin [10]. The *in silico* ADMET study was performed via SwissADME servers to predict the absorption, distribution, metabolism, excretion, and toxicity properties of these compounds [31, 32]. Betanin, isobetainin, and neobetainin deviated from certain drug-likeness criteria, primarily due to higher molecular weights and hydrogen bonding deviations (table 2). Nevertheless, these factors are less critical for topical applications, such as dental disclosing solutions. A higher MW can sometimes be acceptable, particularly if the compound has good solubility and appropriate lipophilicity (logP) to allow for adequate absorption [33]. Interestingly there were many studies reveal that drugs with molecular weight around 500 Da were applied for topical use such as tacrolimus (~800 Da) that has function as topical immunosuppressant [34], cyclosporin derivatives (~1200 Da) that

were applied to treat inflammatory skin disease [35], and hyaluronan (500–1200 kDa) used to attenuate inflammatory and neuropathic pain [36]. All betalains demonstrated high solubility, supporting their formulation into aqueous dental solution. Their low skin permeability, indicated by negative log Kp values, enhances their suitability for topical use, while minimal blood-brain barrier permeability reduces potential side effects in central neural systems (table 2). Importantly, none of the betalains interact with CYP enzymes, indicating a low likelihood of drug-drug interactions. Moderate synthetic accessibility scores in the range of 4.05–6.08 suggest reasonable feasibility for production. Similar to these findings, studies have demonstrated the favorable pharmacokinetics and drug-likeness of betalains contribute to their inhibition *S. mutans* adhesion on *in vitro* studies [14]. In tuberculosis drug development, the favorable pharmacokinetics and drug-likeness of decoquinatone RMB041 offer promising insights, particularly regarding its oral bioavailability, as demonstrated through an *in silico* approach. The findings presented in this study provide a valuable foundation for designing drugs that not only have the potential to shorten the tuberculosis treatment regimen but also minimize adverse effects, thereby improving patient outcome [37]. Other studies revealed that *in silico* and *in vitro* results demonstrate strong concordance in developing natural anthelmintic therapies. *In silico* results revealed samaragenin A and samaragenin B, active compounds in *Syzygium aqueum* as a potent inhibitor of ATP-dependent 6-phosphofructokinase, showing high binding affinities and favorable pharmacokinetics profile. It's aligned with *in vitro* studies, where *S. aqueum* extract significantly affected worm paralysis and mortality [38].

Toxicology study

The toxicological analysis of betalains from *B. vulgaris* showed a range of safety profiles for the ten molecules evaluated, as detailed in table 2. The mutagenicity scores varied from 25% to 61%, with neobetanin having the highest score. Carcinogenicity scores were between 26% and 61%, with betanin and neobetanin also showing elevated values. Hepatotoxicity, an important factor in safety assessment, ranged from 35% to 58%, with betanin and isoleucine-betaxanthin recording the highest levels. Regarding endocrine disruption, both estrogenicity and androgenicity scores were below 37%, suggesting a relatively low risk. Acute oral toxicity ranged from 16% to 39%, while skin and eye irritation scores were moderate, mostly under 25%. Notably, 15-decarboxybetanidin and valine-betaxanthin had lower scores across most toxicity parameters, indicating a safer toxicological profile. The toxicological analysis highlights the diverse safety profiles of betalains, pointing out their

potential risks and benefits for topical applications. Betanin and neobetanin showed favorable pharmacokinetics and solubility but also higher mutagenicity and hepatotoxicity scores, indicating the need for further experimental validation before considering therapeutic uses. In contrast, compounds like 15-decarboxybetanidin and valine-betaxanthin exhibited lower toxicity scores across most parameters, making them safer options for dental disclosing solutions or other topical formulations. The generally low estrogenicity and androgenicity scores for all molecules suggest minimal potential for endocrine disruption, which is beneficial for oral health applications. Although hepatotoxicity scores for several betalains are moderately high (e. g., 57% for betanin and 56% for neobetanin), this risk is less significant for topical applications due to limited systemic absorption, as indicated by the low skin permeability (negative log).

The results of the toxicological analysis are presented in table 2. The results provided insights into safety profiles, which showed low overall mutagenicity, carcinogenicity, estrogenicity, and androgenicity scores, thus within relatively safe profile scores. However, the hepatotoxicity scores have ranged between 55 and 59%, indicating a potential risk, though at a moderate level, to which attention should be given in cases of accidental ingestion or upon long exposure. This aspect, however, needs further deliberation in the context of experimental studies regarding safety, especially for a product to be used within the oral cavity. The results indicate the promising potential of beetroot extract, especially its betalain content, for developing a dental plaque-disclosing solution. Major betalains, like betanin and isobetanin, exhibit proper solubility for inclusion in such a solution. Additionally, their vivid coloring properties make them excellent candidates for effectively staining dental plaque. While some compounds may face pharmacokinetic challenges in drug development, this is less concerning for a topical application like a disclosing solution, where absorption and bioavailability are not critical. The antioxidant and antimicrobial properties of betalains could also offer added benefits by helping detect plaque and providing mild antimicrobial activity, potentially reducing bacterial load, including *S. mutans*, a key player in dental plaque formation. Minimal skin and eye irritation further enhance the safety profile of betalain-based solutions for oral care. The toxicology properties correspond closely with observed *in vitro* antibacterial efficacy, indicating a strong alignment between computational predictions and biological outcomes [39]. In the same way, decoquinatone's *in silico* toxicology profile supports its demonstrated *in vitro* activity, highlighting its potential as an effective oral anti-tuberculosis agent with minimal adverse effects [37].

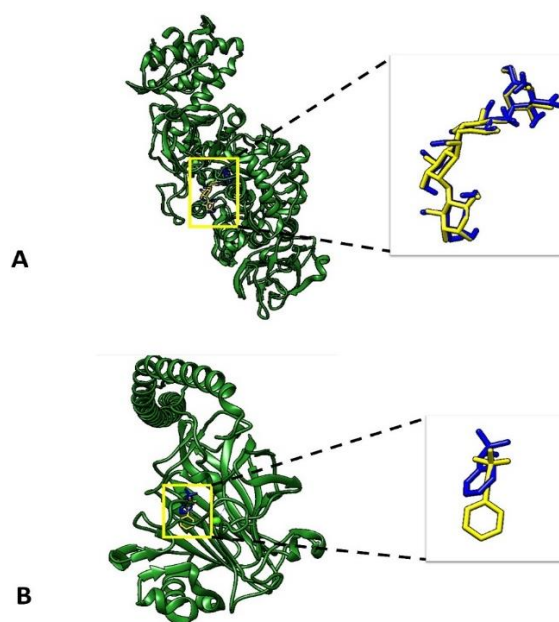


Fig. 5: Docking validation to compare the geometry of the native ligand (yellow) and docked native ligand (blue). A Glucosyltransferase and B Ag I/II

Table 2: Physicochemical properties, pharmacokinetics and toxicological analysis of betalains from *B. vulgaris*

Physicochemical properties										
	Betanin (1)	Isobetanin (2)	Neobetani n (3)	15-Decarboxyb etanidin (4)	17-Decarboxyisob etanidin (5)	Vulgaxanthin I (6)	Leucin betaxanthin_vulgaxanthin IV (7)	Valine-betaxanthin (8)	γ -Aminobutric acid-betaxanthin (9)	Isoleucin - betaxanthin (10)
Molecular weights (Da)	550.47	546.44	548.45	344.32	344.32	339.30	324.33	310.30	296.28	324.33
H-bond acceptors	14	14	14	7	7	8	7	7	7	7
H-bond donors	8	8	8	4	4	5	4	4	4	4
Fraction Csp3	0.42	0.25	0.33	0.24	0.24	0.36	0.47	0.43	0.38	0.47
Rotatable bonds	8	8	8	4	4	8	7	6	7	7
Molar refractivity	135.13	134.46	130.51	96.43	96.43	84.97	86.87	82.07	77.26	86.87
TPSA (Å ²)	247.11	247.11	247.64	130.66	130.66	179.38	136.29	136.29	136.29	136.29
Consensus Log-P	-1.54	-1.42	-1.48	0.76	0.54	-1.31	0.32	0.06	-0.43	0.33
ESOL Log-S	-2.05	-1.75	-2.83	-2.37	-2.27	-0.66	-2.20	-1.95	-1.00	-2.20
ESOL class	Soluble	Very soluble	soluble	Soluble	Soluble	Very soluble	Soluble	Very soluble	Very soluble	Soluble
Lipinski violation	3	3	3	0	0	0	0	0	0	0
Ghose violation	3	2	3	0	0	1	0	0	0	0
Veber violation	1	1	1	0	0	1	0	0	0	0
Egan violation	1	1	1	0	0	1	1	1	1	1
Muegge violation	3	3	3	0	0	1	0	0	0	0
Bioavailability score	0.11	0.11	0.11	0.56	0.56	0.11	0.56	0.56	0.56	0.56
Synthetic accessibility	6.03	6.08	5.37	4.05	4.29	4.62	4.69	4.59	4.14	4.88
Pharmacokinetics										
GI Absorption	Low	Low	Low	High	High	Low	High	High	High	High
BBB permeant	No	No	No	No	No	No	No	No	N	No
Pgp substrate	No	Yes	No	No	No	No	No	No	No	No
Cyp 1A2 inhibitor	No	No	No	No	No	No	No	No	No	No
Cyp 2C19 inhibitor	No	No	No	No	No	No	No	No	No	No
Cyp 2C9 inhibitor	No	No	No	No	No	No	No	No	No	No
Cyp 2D6 inhibitor	No	No	No	No	No	No	No	No	No	No
Cyp 3A4 inhibitor	No	No	No	No	No	No	No	No	No	No
Log-Kp (skin permeation (cm/s))	-10.55	-10.70	-9.77	-7.86	-7.97	-9.22	-7.36	-7.53	-8.35	-7.36
Toxicological Scores (%)										
Mutagenicity	33	33	38	26	35	31	30	27	25	31
Carcinogenicity	42	42	61	41	38	40	38	35	26	38
Hepatotoxicity	57	57	56	54	55	59	56	56	55	57
Estrogenicity	28	28	28	35	36	4	3	3	6	3
Androgenicity	16	16	17	17	15	8	8	7	7	8
Acute Oral toxicity	33	33	31	33	39	19	16	17	20	18
Skin irritation	6	6	7	11	11	18	26	26	24	23
Eye irritation	8	8	4	22	13	12	17	17	23	18

Binding interactions with glucosyltransferase and Ag I/II

The docking validation process employed Root mean Square Deviation (RMSD) values to compare re-docked native ligand to the position of native ligand on the crystal structure of glucosyltransferase and Ag I/II proteins. The RMSD values for glucosyltransferase and Ag I/II were 0.449 Å and 3.790 Å, respectively (table 3 and fig. 5), indicating robust docking accuracy. RMSD < 2 Å is typically regarded as an outstanding re-docking performance, whereas RMSD < 4 Å is regarded as acceptable [26, 27]. Ensuring that the docking protocol can reliably predict ligand binding conformations. Such validation is crucial in molecular docking studies to ensure the credibility of subsequent interaction and binding affinity analyses. The low RMSD values validate the

reliability of the docking method and its ability to reproduce native ligand conformations accurately. This accuracy is pivotal when studying the interactions of *B. vulgaris* (beetroot)-derived compounds with target proteins associated with biofilm formation, particularly *S. mutans* proteins such as glucosyltransferase and Ag I/II. Comparable studies, like those analyzing quorum-sensing inhibitors for biofilm disruption in *Pseudomonas aeruginosa* [40], have similarly emphasized the importance of docking validation for identifying promising compounds with anti-biofilm potential. Furthermore, Martins *et al.* (2021) highlight that precise docking outcomes facilitate identifying key ligand-receptor interactions, a necessary step before molecular dynamics simulations and experimental validations [38, 41]. In this study, validated docking methods provide the foundation for subsequent analysis of beetroot compounds.

Table 3: RMSD value of re-docked native ligand compared to the position of native ligand on the crystal structure to validate the docking method

No	Compared ligand	RMSD (Å)
1	Re-docked native ligand of glucosyltransferase	0.449
2	Re-docked native ligand Ag I/II	3.790

Following those results, the molecular docking binding affinity of bioactive compounds betalains from *B. vulgaris* with glucosyltransferase (fig. 6A) and Ag I/II (fig. 6B), key proteins

involved in *S. mutans* biofilm formation, were shown in table 4. Isobetanin was found to have the highest binding affinity with glucosyltransferase (-10.427 kcal/mol) and Ag I/II (-10.893

kcal/mol), followed closely by betanin and neobetain with -8.842 and -9.324 kcal/mol for glucosyltransferase, and neobetain and betanin with -10.116 and -9.878 kcal/mol for Ag I/II. These values depict strong inhibitory potential in contrast to erythrosine (-7.945 and -8.273 kcal/mol), erythromycin (-7.634 and -7.471 kcal/mol) and native ligand phenylmethanesulfonic acid (-5.278 kcal/mol for Ag I/II). The reference compound alpha-acarbose exhibited maximum affinity in the glucosyltransferase-active site at an interaction energy of -14.561 kcal/mol, hence validating the accuracy of the docking.

The interaction is observed in tables 5 and 6. It describes specific hydrogen and hydrophobic interactions involving active site residues between the ligands and target proteins, including GLU515, ASP477, and ASP588 in glucosyltransferase and SER697, GLU706, and ARG824 in Ag I/II (fig. 7), that provide strong evidence for its disrupting biofilm formation. Other pure compounds, such as betanin and isobetain, also showed excellent interactions with these active sites, contributing to their high binding affinities. Isobetain forming hydrogen bond with glucosyltransferase involving ASP477, SER589, TYR610, ASP909, ASN914, GLN960 and

forming hydrophobic interaction involving LEU433, LEU434, ALA478, ASN481, TRP517, LEU544, HIS587, GLU590, ASN862, LEU908, VAL957, this compound also form attractive interaction with active site GLU515 and ASP588. Meanwhile, comparison with less effective betalain such as vulgaxanthin 1 shows that this betalain does not form hydrogen bond with active site ASP477, GLU515 or ASP588 (data not shown). Furthermore, comparison with native ligand alpha acarbose show that isobetain show similar hydrogen bond with native ligand involving ASP477, ASP909, GLN960 and hydrophobic interaction involving LEU433, LEU434, ALA478, TRP517, HIS587, ASN862, VAL957. This finding support that particular betalain such as isobetain is potential inhibitor of glucosyltransferase. Similar findings were obtained by Knoll *et al.* in 2022 with *Mycobacterium tuberculosis*, where *in silico* docking highlighted important ligand-receptor interactions further validated by *in vitro* assays, showing the translational value of such an *in silico* approach [37]. The dual focus on hydrogen and hydrophobic interactions makes for a strong assertion in making the case of betalains as potent inhibitors of biofilms since effective interaction patterns are crucial for high binding affinities—a concept also reinforced during analysis of natural biofilm disruptors [38, 41].

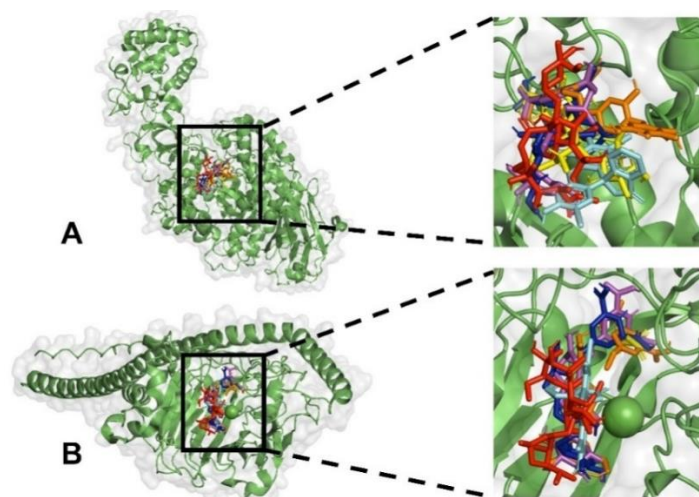


Fig. 6: The binding site of native ligand (yellow), betanin (dark blue), isobetain (orange), neobetain (magenta), erythrosine (cyan), and erythromycin (red). A. glucosyltransferase and B. Ag I/II

Table 4: Molecular docking result of betalains with glycosyltransferase and Ag I/II

No.	Ligand	Binding affinity (kcal/mol)	
		Glucosyltransferase	Ag I/II
1	Native ligand/alpha acarbose	-14.561	
2	Native ligand/phenylmethanesulfonic acid		-5.278
3	Betanin (1)	-9.324	-9.878
4	Isobetain (2)	-10.427	-10.893
5	Neobetain (3)	-8.842	-10.116
6	15-Decarboxybetanidin (4)	-8.077	-8.555
7	17-Decarboxyisobetainidin (5)	-7.937	-8.627
8	vulgaxanthin I (6)	-8.031	-7.687
9	Leucin betaxanthin_vulgaxanthin IV (7)	-7.635	-7.729
10	Valine-betaxanthin (8)	-7.716	-7.405
11	γ-Aminobutric acid-betaxanthin (9)	-7.297	-6.945
12	Isoleucin-betaxanthin (10)	-7.615	-7.271
13	Erythrosine	-7.945	-8.273
14	Erythromycin	-7.634	-7.471

The docking results provide evidence for the betalains, particularly isobetain, betanin and neobetain, as potentially active inhibitors of *S. mutans* biofilm formation. Neobetain, based on its high binding affinity and critical contacts with active site residues, is likely to interfere with the enzymatic functions of glucosyltransferase and with the adhesive properties of Ag I/II (fig. 8). These findings concur with previous studies reporting that natural compounds have a

crucial role in targeting pivotal proteins of biofilm development. Furthermore, the strong binding affinities of betalains, particularly neobetain, isobetain, and betanin, underscore their potential inhibitor of key *S. mutans* biofilm-forming properties. Regarding Suwendar *et al.* (2024), *in silico* results were corroborated by an *in vitro* study, which confirmed inhibitory bacterial biofilm forming [38].

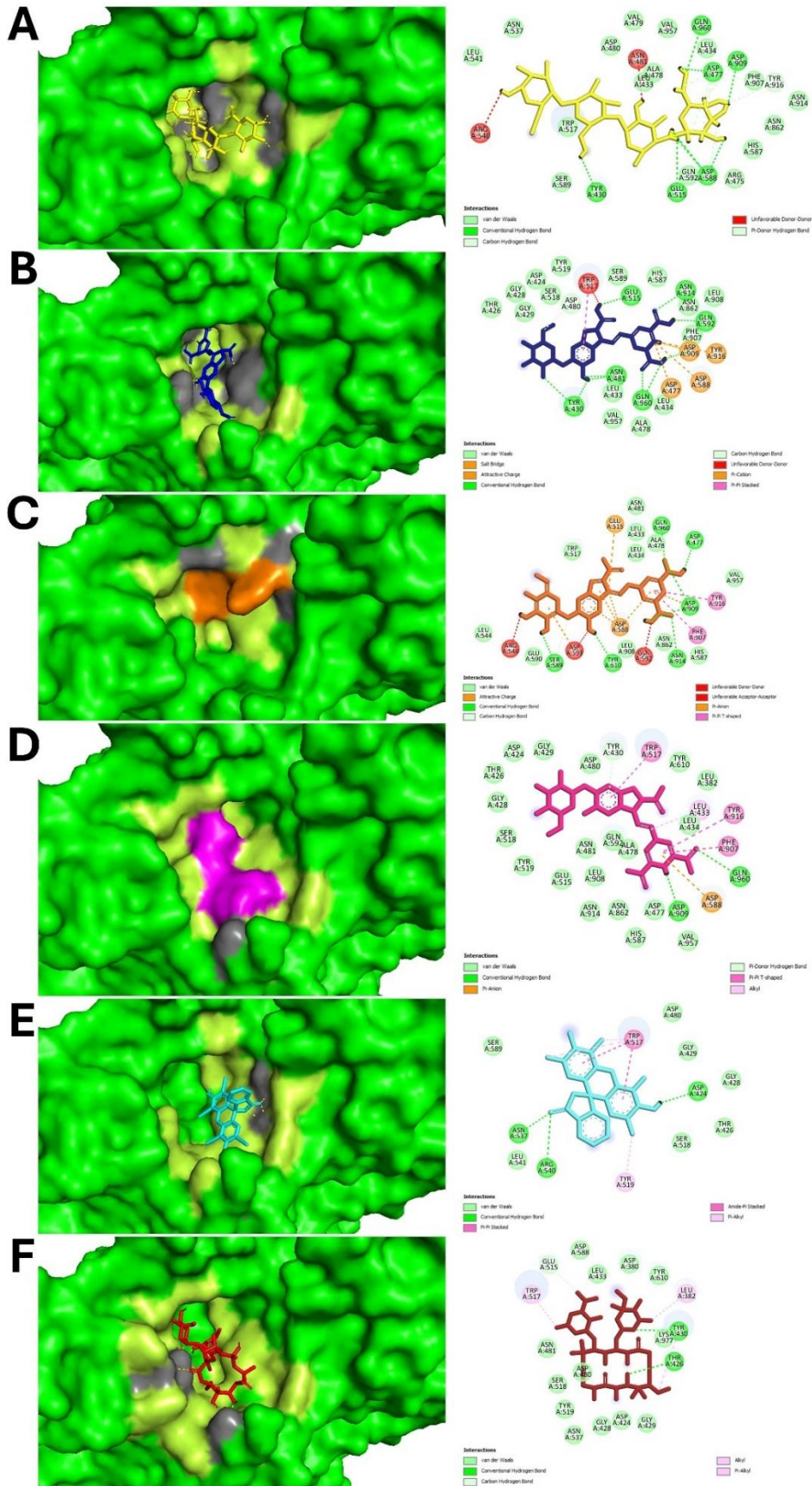


Fig. 7: 3-dimensional (left) and 2-dimensional ligand interaction with glucosyltransferase. A. native ligand (yellow), B. Betanin (dark blue), C. Isobetanin (orange), D. Neobetanin (magenta), E. Erythrosine (cyan) and F. Erythromicin (red). Hydrogen interactions (grey) and hydrophobic interactions (limon) are shown in 3-dimentional fig. (left)

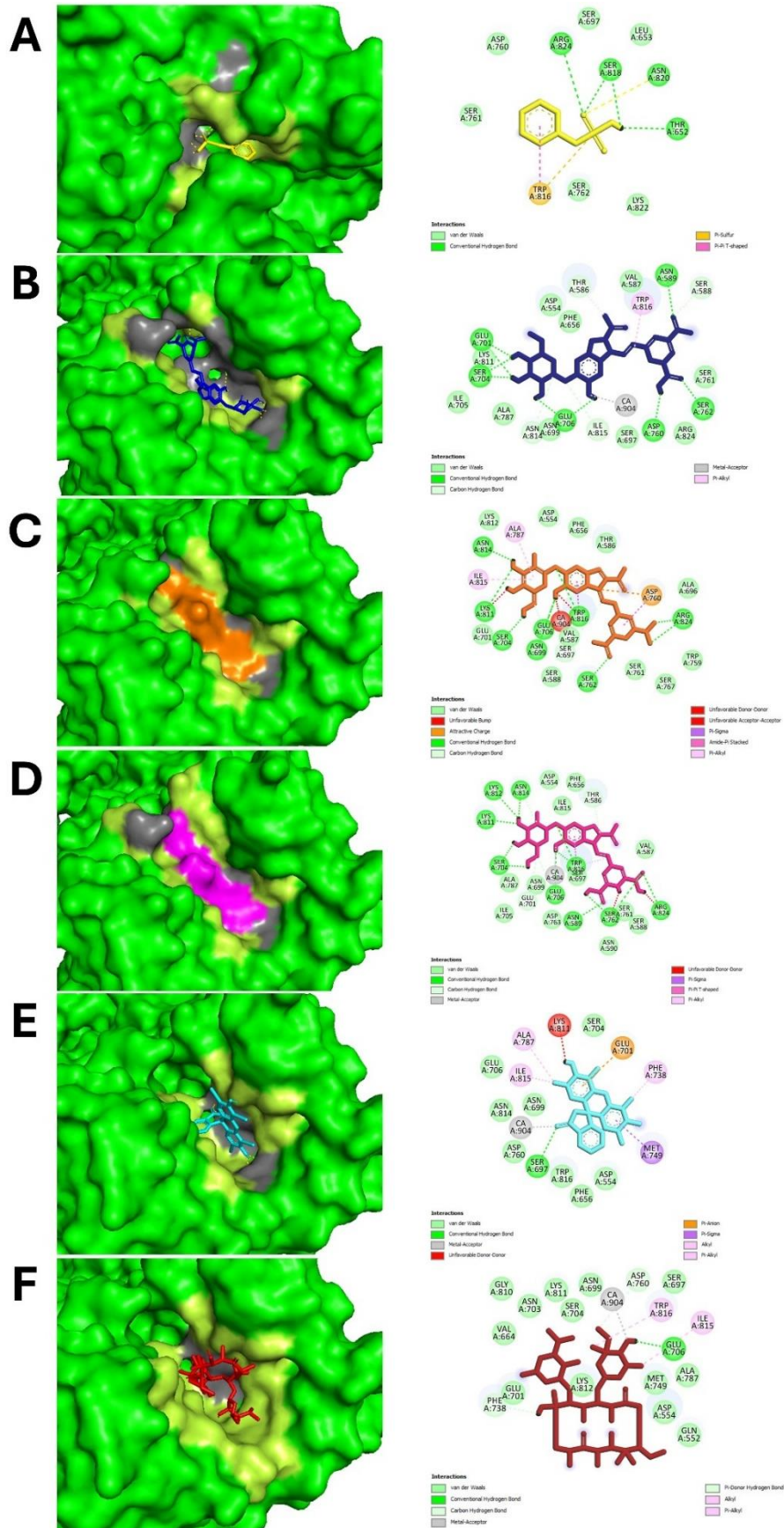


Fig. 8: 3-dimensional (left) and 2-dimensional ligand interaction with Ag I/II. A. native ligand (yellow), B. Betanin (dark blue), C. Isobetainin (orange), D. Neobetainin (magenta), E. Erythrosine (cyan) and F. Erythromycin (red). Hydrogen interactions (grey) and hydrophobic interactions (limon) are shown in 3-dimentional fig. (left)

Compared to erythrosine, the synthetic dye generally used in dental plaque-disclosing solutions, betalains exhibited higher binding affinities and more interactions with essential active sites. It supports the idea of replacing synthetic agents with natural alternatives that offer minimal or no toxicity while maintaining or exceeding efficacy. Additionally, interactions of isobetanin, betanin, and neobetainin with critical residues GLU515 and ASP477 in glucosyltransferase align with studies highlighting the role of these amino acids in inhibiting glucosyltransferase enzymatic activity. Furthermore, the findings suggest that betalains may act as dual-

function agents; in addition to plaque detection enabled by their bright color, they may interfere with microbial biofilm formation. Future studies using molecular dynamics and *in vitro* validation will help confirm the stability and efficacy of the interactions of betalains with proteins under physiological conditions, paving the way for developing safer and more effective dental care formulations. Furthermore, isobetanin, betanin, and neobetainin also have higher binding affinities than erythromycin. Erythromycin is effective antibiotic against *S. mutans*, a type of bacteria that can cause tooth decay [42].

Table 5: Amino acids on glycosyltransferase in-volve in hydrogen and hydrophobic interaction with native ligand, betanin, isobetanin and neobetainin

No.	Ligand	Amino acid residues			
		Hydrogen interaction	Hydrophobic interaction	Salt bridge/Attractive charge/Pi-Cation	Pi-Pi stacked/Alkyl
1	Native ligand	TYR430, ASP477, GLU515, ASP588, PHE907, ASP909, GLN960	LEU433, LEU434, ARG475, ALA478, VAL479, ASP480, TRP517, ASN537, LEU541, HIS587, SER589, GLN592, ASN862, ASN914, TYR916, VAL957		
2	Betanin (1)	TYR430, ASP480, ASN481, GLU515, GLN592, ASP909, ASN914, GLN960	ASP424, THR426, GLY428, GLY429, LEU433, LEU434, ALA478, SER518, TYR519, HIS587, SER589, ASN862, PHE907, LEU908, VAL957	ASP477, ASP588, ASP909, TYR916	TRP517
3	Isobetanin (2)	ASP477, SER589, TYR610, ASP909, ASN914, GLN960	LEU433, LEU434, ALA478, ASN481, TRP517, LEU544, HIS587, GLU590, ASN862, LEU908, VAL957	GLU515, ASP588, ASP593, ASP909	PHE907, TYR916
4	Neobetainin (3)	ASP909, GLN960	LEU382, ASP424, THR426, GLY428, GLY429, LEU434, ASP477, ALA478, ASP480, ASN481, GLU515, SER518, TYR519, HIS587, GLN592, TYR610, ASN862, LEU908, ASN914, VAL957	ASP588	LEU433, TRP517, PHE907, TYR916
5	Erythrosine	ASP424, GLU515, ASN537, ARG540	THR426, GLY428, GLY429, ASP480, SER518, LEU541, SER589		TRP517, TYR519
6	Erythromicin	THR426, TYR430	ASP380, ASP424, GLY428, GLY429, LEU433, ASP480, ASN481, SER518, TYR519, ASN537, ASP588, TYR610, LYS977		LEU382, TRP517

Table 6: Amino acids on Ag I/II in volve in hydrogen and hydrophobic interaction with native ligand, betanin, isobetanin and neobetainin

S. No.	Ligand	Amino acid residues			
		Hydrogen interaction	Hydrophobic interaction	Salt bridge/Attractive charge/Pi-cation	Pi-Pi stacked/Alkyl
1	Native ligand	THR652, SER818, ASN820, ARG824	LEU653, SER697, ASP760, SER761, SER762, LYS822	TRP816	TRP816
2	Betanin (1)	SER588, ASN589, GLU701, SER704, GLU706, ASP760, SER762, ASN814, ILE815	ASP554, VAL587, PHE656, SER697, ASN699, ILE705, SER761, ALA787, LYS811, ARG824		TRP816
3	Isobetanin (2)	ASN699, SER704, GLU706, SER762, LYS811, ASN814, TRP816, ARG824,	ASP554, THR586, VAL587, SER588, PHE656, SER697, ALA696, GLU701, TRP759, SER761, SER767, LYS812	ASP760	ASP760, ALA787, ILE815, TRP816
4	Neobetainin (3)	ASN589, GLU701, SER704, GLU706, SER762, LYS811, LYS812, ASN814, TRP816, ARG824	ASP554, VAL587, SER588, ASN590, PHE656, SER697, ASN699, ILE705, SER761, ASP763, ALA787, ILE815		TRP816
5	Erythrosine	SER697	ASP554, PHE656, ASN699, SER704, GLU706, ASP760, ASN814, TRP816	GLU701	PHE738, MET749, ALA787, ILE815
6	Erythromicin	GLU706, PHE738, ASP760	GLN552, ASP554, VAL664, SER697, ASN699, GLU701, ASN703, SER704, MET749, ALA787, GLY810, LYS811, LYS812		ILE815, TRP816

Antibacterial activity of beetroot extract

The antibacterial test aims to determine the activity of beetroot extract in inhibiting the growth of *S. mutans*, which is the main cause of dental caries. Beetroot extracts from three different areas, betanin, and Amoxicillin showed antibacterial activity by forming a clear zone (fig. 9). Beetroot extract from Magelang had a higher inhibition zone diameter than betanin and extracts from other areas (table 7). However, the measured inhibition zone diameter of the beetroot extract from Magelang was low compared to the inhibition

zone diameter of Amoxicillin. The administration of beetroot extract from Magelang at a concentration of 1000 mg/ml only produced an inhibition zone of 12±3 mm, while the administration of Amoxicillin at a concentration of 0.5 mg/ml showed a much larger inhibition zone of 31.7 mm. Davis and Stout (1971) revealed that the diameter of the inhibition zone can be grouped into four categories, namely weak (<5 mm), moderate (5-10 mm), strong (10-20 mm), and very strong (>20 mm) [43]. However, the results of the inhibition zone measurements on the samples used in the research must be compared with positive controls that have been commercially used

and proven to inhibit bacterial growth. Other results showed that the broth microdilution method of beetroot extracts from Magelang, West Bandung, Ciwidey, and betalain had antibacterial activity against *S. mutans* with MIC values of 125 mg/ml and 250 mg/ml respectively (table 7). The MIC value shows that beetroot extract from Magelang has better antibacterial activity compared to beetroot extract from other areas previous studies have shown that beetroot ethanol extract can inhibit the growth of *S. mutans*, and the inhibitory activity increases with increasing concentration of the extract [44]. Other studies have shown that beetroot ethanol extract can inhibit g-positive bacteria, depending on the concentration [45]. Betalains from beetroot ethanol extract are known to inhibit the growth of *E. fecalis*, the antibacterial activity increases with increasing concentration [46]. However, in this study, the antibacterial activity of betanin inhibiting the growth of *S. mutans*

was lower compared to the ethanol extract of beetroot from Magelang. This is possible because the ethanol extract of beetroot from Magelang not only contains betalain, but also contains other compounds that work synergistically to inhibit the growth of *S. mutans*. Previous studies have shown that beetroot antibacterial activity is possible due to the content of various other active compounds, such as phenols, flavonoids, saponins, and tannins [47, 48]. These active compounds cause various damages to the bacterial structure, thereby inhibiting the metabolism of the bacterial body and ultimately causing bacterial death [49–52]. Beetroot activity in inhibiting *S. mutans* needs to be studied further to validate its effectiveness as an agent that can inhibit the formation of dental caries. Further research also needs to determine the activity of beetroot in inhibiting other bacteria that cause dental caries, besides *S. mutans*.

Table 7: Antibacterial activity of beetroot extract against *S. mutans*

Sample		Diameter of inhibition zone (mm)	MIC (mg/ml)
Beetroot extract	Bandung Barat	6.7±0.6	250.0±0.0
	Ciwidey	6.7±0.6	250.0±0.0
	Magelang	12±3	125.0±0.0
Betanin		6.0± 0.0	250±0.0
Amoxicillin (positive control)		31.7±2.1	976.5x 10 ⁻⁶ ± 0.0

Value showed the mean±SD (n = 3)



Fig. 9: The diameter of inhibition zone from beetroot extract against *S. mutans*. A: Amoxicillin (positive control); BB: beetroot extract from Bandung Barat; C: beetroot extract from Ciwidey; M: beetroot extract from Magelang; Be: Betalain

CONCLUSION

Betalains from *Beta vulgaris* emerge as promising dual-function agents in oral healthcare, uniquely combining vibrant plaque detection with potent biofilm inhibition. Isobetanin, betanin and neobetanin exhibited excellent binding affinities toward glucosyltransferase and Ag I/II, two essential proteins in the biofilm formation of *Streptococcus mutans*, which was confirmed by highly accurate docking results with RMSD values below 4 Å. High solubility, low toxicity, and minimal systemic effects further reinforce their appropriateness for dental disclosing solutions. These findings underpin the potential transformative power of betalains as non-toxic, natural alternatives to synthetic dyes in dental applications, with comprehensive impacts on the development of new, efficient methods and means in diagnostics and prevention of oral health disorders. Further experimental validation will definitely fix their place as one of the cornerstones in next-generation dental formulations. *In vitro* studies have shown that beetroot ethanol extract can inhibit the growth of *S. mutans* better than betalain. This occurs because beetroot ethanol extract contains not only betalains but also other compounds that synergistically and

comprehensively inhibit the growth of *S. mutans*. Further research is needed to investigate the effect of beetroot ethanol extract on inhibiting *S. mutans* biofilm formation, to understand the potential of beetroot as an active ingredient in disclosing solutions. Moreover, future direction will be focused on formulation of disclosing solution. The resulting disclosing solution will be tested in a laboratory to ensure its ability to effectively identify dental plaque on stored biological material in the form of extracted human tooth samples.

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AUTHORS CONTRIBUTIONS

Sofa Fajriah and Yahdiana Harahap contributed resources, conceptualization and supervision, and reviewing and editing of the manuscript; Nur Fitriana and Ilma Fauziah Ma'rif to in silico study analysis; Rifaldi and Fadhila Utari to data curation, validation, and writing original manuscript, Susi Kusumaningrum contributed to UPLC analysis, Bantari Wisynu Kusuma Wardhani contributed to the writing original manuscript editing of the manuscript. Sri Ratna Laksmiastuti contributed to resources, while Reynatha C. A. Pangsidang contributed to manuscript review and editing.

CONFLICTS OF INTERESTS

The authors declare no conflict of interest in this manuscript.

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